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NEWS 6 APR 26 USPATFULL and USPAT2 enhanced with patent
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NEWS 8 APR 28 ENCOMPLIT/ENCOMPLIT2 search fields enhanced
NEWS 9 APR 28 Limits doubled for structure searching in CAS
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NEWS 10 MAY 08 STN Express, Version 8.4, now available
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STN Easy
NEWS 13 MAY 14 DGENE, PCTGEN and USGENE enhanced with increased
limits for exact sequence match searches and
introduction of free HIT display format
NEWS 14 MAY 15 INPADOCDB and INPAFAMDB enhanced with Chinese legal
status data
NEWS 15 MAY 28 CAS databases on STN enhanced with NANO super role in
records back to 1992
NEWS 16 JUN 01 CAS REGISTRY Source of Registration (SR) searching
enhanced on STN
NEWS 17 JUN 26 NUTRACEUT and PHARMAML no longer updated
NEWS 18 JUN 29 IMSCOPROFILE now reloaded monthly
NEWS 19 JUN 29 EFPULL adds Simultaneous Left and Right Truncation
(SLART) to AB, MCLM, and TI fields
NEWS 20 JUL 09 PATDPFULL adds Simultaneous Left and Right
Truncation (SLART) to AB, CLM, MCLM, and TI fields

NEWS EXPRESS MAY 26 09 CURRENT WINDOWS VERSION IS V8.4,
AND CURRENT DISCOVER FILE IS DATED 06 APRIL 2009.

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=> file reg		
COST IN U.S. DOLLARS	SINCE FILE	TOTAL
	ENTRY	SESSION
FULL ESTIMATED COST	0.66	0.66

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STRUCTURE FILE UPDATES: 8 JUL 2009 HIGHEST RN 1161500-61-3
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<http://www.cas.org/support/stngen/stdoc/properties.html>

=>

Uploading C:\Program Files\Stnexp\Queries\10580384.str

L1 STRUCTURE UPLOADED

=> d l1

L1 HAS NO ANSWERS

L1 STR



Structure attributes must be viewed using STN Express query preparation.

=> s l1

SAMPLE SEARCH INITIATED 13:55:21 FILE 'REGISTRY'

SAMPLE SCREEN SEARCH COMPLETED - 13525 TO ITERATE

14.8% PROCESSED 2000 ITERATIONS 50 ANSWERS
INCOMPLETE SEARCH (SYSTEM LIMIT EXCEEDED)
SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**

BATCH **COMPLETE**

PROJECTED ITERATIONS: 263531 TO 277469

PROJECTED ANSWERS: 90284 TO 98524

L2 50 SEA SSS SAM L1

=> s l1 full

FULL SEARCH INITIATED 13:55:25 FILE 'REGISTRY'

FULL SCREEN SEARCH COMPLETED - 273365 TO ITERATE

100.0% PROCESSED 273365 ITERATIONS 94871 ANSWERS
SEARCH TIME: 00.00.02

L3 94871 SEA SSS FUL L1

=> file caplus
COST IN U.S. DOLLARS

SINCE FILE	TOTAL
ENTRY	SESSION
185.88	186.54

FULL ESTIMATED COST

FILE 'CAPLUS' ENTERED AT 13:55:31 ON 09 JUL 2009
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FILE COVERS 1907 - 9 Jul 2009 VOL 151 ISS 2
FILE LAST UPDATED: 8 Jul 2009 (20090708/ED)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Apr 2009
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Apr 2009

CPlus now includes complete International Patent Classification (IPC) reclassification data for the second quarter of 2009.

CAS Information Use Policies apply and are available at:

<http://www.cas.org/legal/infopolicy.html>

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s l3

L4 31119 L3

=> s l4 and acrylic

316082 ACRYLIC
L5 1508 L4 AND ACRYLIC

=> s l4 and methacrylic

88473 METHACRYLIC
L6 503 L4 AND METHACRYLIC

=> s l4 and (meth)acrylic

MISSING OPERATOR METH)ACRYLIC

The search profile that was entered contains terms or nested terms that are not separated by a logical operator.

=> s l5 or l6

L7 1685 L5 OR L6

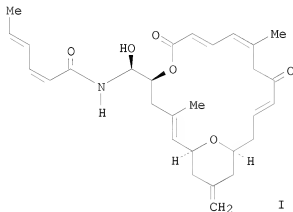
=> s l7 and esterification

108157 ESTERIFICATION
L8 25 L7 AND ESTERIFICATION

=> d l8 ibib abs 1-

YOU HAVE REQUESTED DATA FROM 25 ANSWERS - CONTINUE? Y/(N):y

L8 ANSWER 1 OF 25 CAPLUS COPYRIGHT 2009 ACS ON STN
ACCESSION NUMBER: 2008:1032796 CAPLUS
DOCUMENT NUMBER: 150:374168
TITLE: Studies toward the synthesis of (-)-zampanolide:
preparation of the macrocyclic core
AUTHOR(S): Troast, Dawn M.; Yuan, Jiayi; Porco, John A., Jr.
CORPORATE SOURCE: Department of Chemistry, Center for Chemical
Methodology and Library Development (CMLD-BU), Boston
University, Boston, MA, 02215, USA
SOURCE: Advanced Synthesis & Catalysis (2008), 350(11+12),
1701-1711
CODEN: ASCAF7; ISSN: 1615-4150
PUBLISHER: Wiley-VCH Verlag GmbH & Co. KGaA
DOCUMENT TYPE: Journal
LANGUAGE: English
OTHER SOURCE(S): CASREACT 150:374168
GI



AB Studies towards the synthesis of the macrocyclic core of (-)-zampanolide (I) are reported. The synthetic approach features a one-pot reduction/vinylogous aldol reaction for construction of the C-15-C-20 fragment, an intramol. silyl-modified Sakurai (ISMS) reaction for construction of the 2,6-cis-disubstituted exo-methylene pyran subunit, and use of an sp²-sp³ Stille reaction for macrocyclization.

REFERENCE COUNT: 102 THERE ARE 102 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 2 OF 25 CAPLUS COPYRIGHT 2009 ACS ON STN
ACCESSION NUMBER: 2006:718802 CAPLUS
DOCUMENT NUMBER: 145:335974
TITLE: SuperQuat 5,5-dimethyl-4-iso-propyloxazolidin-2-one as a mimic of Evans 4-tert-butylloxazolidin-2-one
AUTHOR(S): Bull, Steven D.; Davies, Stephen G.; Garner, A. Christopher; Kruchinin, Dennis; Key, Min-Suk; Roberts, Paul M.; Savory, Edward D.; Smith, Andrew D.; Thomson, James E.
CORPORATE SOURCE: The Department of Organic Chemistry, Chemistry Research Laboratory, University of Oxford, Oxford, OX1 3TA, UK

SOURCE: Organic & Biomolecular Chemistry (2006), 4(15),
2945-2964
CODEN: OBCRAK; ISSN: 1477-0520
PUBLISHER: Royal Society of Chemistry
DOCUMENT TYPE: Journal
LANGUAGE: English
OTHER SOURCE(S): CASREACT 145:335974

AB The incorporation of a gem-di-Me group at the 5-position of a chiral oxazolidinone biases the conformation of the adjacent C(4)-stereodirecting group such that the gem-dimethyl-4-iso-Pr combination mimics a C(4)-tert-Bu group, providing higher levels of stereocontrol than a simple 4-isopropylloxazolidinone. The stereoselectivities of alkylation, esterification (O-acylation), Diels-Alder, and alkene oxidative acetalization reactions of acyloxazolidinones with either a 4-iso-Pr or 4-tert-Bu group and either possessing or lacking gem-5,5-dimethyl groups are compared.

REFERENCE COUNT: 40 THERE ARE 40 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 3 OF 25 CAPLUS COPYRIGHT 2009 ACS ON STN

ACCESSION NUMBER: 2005:822461 CAPLUS
DOCUMENT NUMBER: 143:212623
TITLE: Crosslinking of compounds having hydroxy groups bonded to aromatic rings and the crosslinked products
INVENTOR(S): Morita, Takehiko; Tsujino, Naoshi
PATENT ASSIGNEE(S): Nippon Shokubai Co., Ltd., Japan
SOURCE: Jpn. Kokai Tokkyo Koho, 13 pp.
CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2005220221	A	20050818	JP 2004-29250	20040205
PRIORITY APPLN. INFO.:			JP 2004-29250	20040205

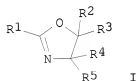
AB Aromatic compds. having ≥ 1 OH and (meth)acryloyloxy or (meth)acryloyloxyalkoxy groups on the same aromatic ring are allowed to react with compds. having oxazoline, epoxy, and/or isocyanate groups. Thus, a mixture of 2-isopropenyl-2-oxazoline 20, Me methacrylate 50, Bu acrylate 20, 2-(2-acryloyloxyethoxy)phenol (preparation described) 10, and AIBN 1.5 parts was added dropwise to EtOAc at 80° for 4 h, heated to 90° for 2.5 h, and cooled to give a solution of a prepolymer having phenolic OH and oxazoline groups, which was mixed with 0.1 phr di-Ph phosphite, applied on a glass sheet, and crosslinked at 180° for 1 h to form a water-resistant coating.

L8 ANSWER 4 OF 25 CAPLUS COPYRIGHT 2009 ACS ON STN

ACCESSION NUMBER: 2005:472105 CAPLUS
DOCUMENT NUMBER: 143:8161
TITLE: Method for the esterification of alcohols with olefinically unsaturated carboxylic acids
INVENTOR(S): Glos, Martin
PATENT ASSIGNEE(S): Clariant G.m.b.H., Germany
SOURCE: PCT Int. Appl., 20 pp.
CODEN: PIXXD2
DOCUMENT TYPE: Patent
LANGUAGE: German
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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WO 2005049544	A1	20050602	WO 2004-EP12790	20041111
W:	AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW			
RW:	BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG			
DE 10354652	A1	20050707	DE 2003-10354652	20031122
AU 2004291298	A1	20050602	AU 2004-291298	20041111
CA 2546819	A1	20050602	CA 2004-2546819	20041111
EP 1687251	A1	20060809	EP 2004-797820	20041111
EP 1687251	B1	20090318		
R:	AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK, IS			
BR 2004016734	A	20070116	BR 2004-16734	20041111
CN 1902155	A	20070124	CN 2004-80030595	20041111
JP 2007511561	T	20070510	JP 2006-540260	20041111
ES 2320227	T3	20090520	ES 2004-797820	20041111
MX 2006005674	A	20060817	MX 2006-5674	20060519
US 20070149803	A1	20070628	US 2006-580384	20060522
PRIORITY APPLN. INFO.:			DE 2003-10354652	A 20031122
			WO 2004-EP12790	W 20041111
OTHER SOURCE(S):	MARPAT 143:8161			
GI				



AB The invention relates to a method for producing esters from alcs. and olefinically unsatd. carboxylic acids by reacting an alc. with an olefinically unsatd. carboxylic acid or a reactive derivative thereof, in the presence of between 1 ppm and 1 weight % oxazoline I in relation to the weight of the reaction mixture of alc. and olefinically unsatd. carboxylic acid/carboxylic acid derivative, R1, R2, R3, R4 and R5 representing hydrogen or branched, linear, cyclical, saturated or unsatd. hydrocarbon radicals containing up to 25 C atoms that can be substituted by heteroatoms, and R1, R2, R3, R4 and R5 being the same or different to prevent formation of polymer on surfaces in reactor.

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 5 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN
 ACCESSION NUMBER: 2004:651161 CAPLUS
 DOCUMENT NUMBER: 141:175121
 TITLE: Transparent scratch-resistant laminated polyester film
 INVENTOR(S): Tojo, Mitsumine
 PATENT ASSIGNEE(S): Teijin Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 13 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent

LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2004223938	A	20040812	JP 2003-15745	20030124
PRIORITY APPLN. INFO.:			JP 2003-15745	20030124

AB The laminated film, useful for an optical film, has thickness 20-300 μ m and comprises a polyester substrate layer having at least thereon a coating layer, wherein (i) the substrate layer is substantially free from particles and the coating layer contains inert particles with mean particle diameter 10-150 nm and (ii) the substrate layer contains cyclic trimer $\leq 0.8\%$ and concentration of terminal carboxyl group of the polymer ≤ 30 equiv/106 g. Thus, poly(ethylene terephthalate) with intrinsic viscosity 0.60, prepared by esterification, GeO₂-catalyzed polycondensation, and solid-phase polymerization of bis(β -hydroxyethyl) terephthalate, terephthalic acid, and ethylene glycol at parts ratio 100:65:29, was brought into contact with hot water (90°), dried, extruded at 295°, quenched on a drum to give an undrawn film, and stretched 3.6-folds at 100° in the machine direction to give a uniaxially drawn substrate film. A coating containing 78 parts 90:5:5:90:50 (mol%) di-Me 2,6-naphthalenedicarboxylate-dimethyl isophthalate-dimethyl 5-sodiosulfisophthalate-ethylene glycol-diethylene glycol copolymer, 15 parts 10:70:5:15 (mol%) Me methacrylate-2-isopropenyl-2-oxazoline-poly(ethylene oxide) methacrylate-acrylamide copolymer, 2 parts SiO₂ (Cataloid Si 80p), and 5 parts Naroacty N 70 (polyoxyethylene lauryl ether) was applied on the substrate film, stretched 3.8-folds in the transverse direction at 110°, and set at 215° to give a 100- μ m thick film with total light transmittance 92%, haze $\leq 0.8\%$, and good resistances to scratch and heat.

L8 ANSWER 6 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN
 ACCESSION NUMBER: 2001:417041 CAPLUS
 DOCUMENT NUMBER: 135:34551
 TITLE: Hydrophilic hydrogel-forming polymers comprising 1,4- α -D-glycosidic bonds
 INVENTOR(S): Engelhardt, Friedrich; Frenz, Volker; Herfert, Norbert; Riegel, Ulrich; Weismantel, Matthias
 PATENT ASSIGNEE(S): Basf Aktiengesellschaft, Germany
 SOURCE: PCT Int. Appl., 28 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: German
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2001040335	A1	20010607	WO 2000-EP11276	20001115
W: BR, CA, JP, MX				
RW: AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR				
PRIORITY APPLN. INFO.:			US 1999-450948	A 19991129

AB The title polymers comprising 1,4- α -D-glycosidic bonds and acid groups and/or their alkali or ammonium salts are post-crosslinked on the surface and have (a) a centrifuge retention capacity (CRC) parameter of 1.10 and (b) a vertical absorption (test description given) of <12 g/g. The polymers have good absorptive power and retention of liqs. and are useful for the manufacture of diapers and related articles. For example, redox polymerization of acrylic acid and methylenebisacrylamide copolymer in the presence of starch allyl ether as crosslinker (preparation given) gave a solid gel which was comminuted, partly neutralized with aqueous NaOH, dried

and surface-modified with diglycidyl propylphosphonate and ethylene glycol diglycidyl ether to give a title hydrogel having CRC parameter 1.293 and vertical absorption 12.5 g/g.

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 7 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2000:153666 CAPLUS

DOCUMENT NUMBER: 132:279065

TITLE: Ruthenium Carbene Complexes with N,N'-Bis(mesityl)imidazol-2-ylidene Ligands: RCM Catalysts of Extended Scope

AUTHOR(S): Fuerstner, Alois; Thiel, Oliver R.; Ackermann, Lutz; Schanz, Hans-Joerg; Nolan, Steven P.

CORPORATE SOURCE: Max-Planck-Institut fuer Kohlenforschung, Muelheim/Ruhr, D-45470, Germany

SOURCE: Journal of Organic Chemistry (2000), 65(7), 2204-2207 CODEN: JOCEAH; ISSN: 0022-3263

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 132:279065

AB The Ru carbene complexes Cl₂Ru(PCy₃)L(L') 3a,b (L = N,N'-dimesityl-2,3-dihydro-1H-imidazol-2-ylidene; L' = benzylidene, 3-phenylindenylidene) constitute excellent precatalysts for ring-closing metathesis (RCM) reactions allowing the formation of tri- and tetrasubstituted cycloalkenes (e.g. 3,4-dimethyl-3-cyclopentene-1,1-dicarboxylic acid di-Et ester). They also appear to annulations that are beyond the scope of the standard Grubbs carbene Cl₂Ru(PCy₃)₂(:CHPh) (e.g. 1,2,3,4,5,6-hexahydroindene-3a-carboxylic acid Me ester) as well as to ring-closing reactions of acrylic acid derivs. even if the resulting α,β -unsatd. lactones (or lactams) are tri- or tetrasubstituted (e.g. 5-ethyl-3,4-dimethyl-2(5H)-furanone). The reactivity of 3a is highly dependent on the reaction medium: particularly high reaction rates are observed in toluene, although this solvent also leads to an increased tendency of the catalyst to isomerize the double bonds of the substrates.

REFERENCE COUNT: 52 THERE ARE 52 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 8 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1996:243094 CAPLUS

DOCUMENT NUMBER: 125:34317

ORIGINAL REFERENCE NO.: 125:6721a,6724a

TITLE: FTIR spectroscopic studies on the interfacial reactions of oxazoline-functionalized polymers

AUTHOR(S): Schaefer, R.; Kressler, J.; Muelhaupt, R.

CORPORATE SOURCE: Inst. Makromol. Chem., Albert-Ludwigs-Univ., Freiburg, D-79104, Germany

SOURCE: Acta Polymerica (1996), 47(4), 170-6

CODEN: ACPODY; ISSN: 0323-7648

PUBLISHER: VCH

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The interfacial reaction of immiscible polymeric systems containing oxazoline and carboxylic acid groups, resp., was studied by FTIR spectroscopy and FTIR microscopy. A butadiene-acrylonitrile copolymer rubber, where the nitrile groups were partially converted to oxazoline groups, was thermally annealed in a 2-layer specimen with ethene-methacrylic acid copolymer (PEM). The formation of the ester amide in the interface was measured quant. by FTIR difference spectroscopy. Evaluating the formation of the amide I band led to reaction rate consts. of 0.0125-0.022 s⁻¹ at 170-230°. This system was immiscible because there is no

interdiffusion across the interface. Also, the formation of an ester amide in the interfacial reaction of the system
 PEM/bis(1,3-oxazoline-2-yl)-terminated di-Me siloxane was monitored qual.
 by FTIR spectroscopy.

L8 ANSWER 9 OF 25 CAPLUS COPYRIGHT 2009 ACS ON STN

ACCESSION NUMBER: 1994:108707 CAPLUS
 DOCUMENT NUMBER: 120:108707
 ORIGINAL REFERENCE NO.: 120:19211a,19214a
 TITLE: Methacryloxyethyl vinyl carbonate as novel UV-curable crosslinking agent for acrylic, vinyl, and styrenic hydrophilic monomers
 INVENTOR(S): Lai, Yu Chin
 PATENT ASSIGNEE(S): Bausch and Lomb Inc., USA
 SOURCE: PCT Int. Appl., 101 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 2
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 9309084	A1	19930513	WO 1992-US9539	19921103
W: AU, BR, CA, JP, KR				
RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, SE				
US 5310779	A	19940510	US 1992-922452	19920730
AU 9230664	A	19930607	AU 1992-30664	19921103
AU 664401	B2	19951116		
EP 611367	A1	19940824	EP 1992-924307	19921103
EP 611367	B1	19970625		
R: BE, DE, ES, FR, GB, GR, IE, IT, NL, SE				
JP 07500865	T	19950126	JP 1992-508729	19921103
BR 9206886	A	19950411	BR 1992-6886	19921103
CA 2121577	C	20000620	CA 1992-2121577	19921103
JP 3490436	B2	20040126	JP 1993-508729	19921103
IN 186331	A1	20010811	IN 1992-DE1017	19921105
PRIORITY APPLN. INFO.:				
			US 1991-788071	A 19911105
			US 1992-884481	A 19920515
			US 1992-922452	A 19920730
			WO 1992-US9539	A 19921103

AB UV-cured hydrogels prepared by polymerizing ≥ 1 (meth)acrylate- or (meth)acrylamide-capped urethane-siloxane macromer of specified structure, ≥ 1 vinyl monomer, e.g., N-vinylpyrrolidone, and ≥ 1 crosslinking agent of specified structure, e.g. methacryloxyethyl vinyl carbonate (I), -carbamate, or -urea, 4-vinylphenyl vinyl urea, etc., are useful for the manufacture of biomedical devices, especially contact lenses.

Thus, a mixture of α,ω -bis(methacryloxybutyl)polysiloxane 13, 3-methacryloxypropyl tris(trimethylsiloxy)silane 47, N-vinylpyrrolidone 40, I (preparation from CH₂:CMeCO₂CH₂CH₂OH and ClCO₂CH:CH₂ given) 0.1, Darocur 1173 0.2, and benzoin Me ether 0.2 parts in hexanol was cast onto a glass plate, UV-cured, extracted, and equilibrated in buffered saline at pH 7.4 to give a film containing 23% H₂O and having modulus 369 g/mm², contact angle 20°, and O permeability 114 Dk units.

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 10 OF 25 CAPLUS COPYRIGHT 2009 ACS ON STN

ACCESSION NUMBER: 1993:628125 CAPLUS
 DOCUMENT NUMBER: 119:228125
 ORIGINAL REFERENCE NO.: 119:40695a,40698a
 TITLE: Low loss high numerical aperture cladded optical

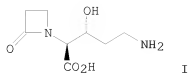
INVENTOR(S): fibers
 Babirad, Stefan A.; Kuczma, Andrew S.; Savu, Patricia M.
 PATENT ASSIGNEE(S): Minnesota Mining and Manufacturing Co., USA
 SOURCE: PCT Int. Appl., 35 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 9304132	A1	19930304	WO 1992-US5976	19920716
W: CA, JP, KR				
RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LU, MC, NL, SE				
US 5239026	A	19930824	US 1991-750092	19910826
EP 600985	A1	19940615	EP 1992-917902	19920716
EP 600985	B1	19970115		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, MC, NL, SE				
JP 06510316	T	19941117	JP 1992-504296	19920716
JP 2756188	B2	19980525		
ES 2096098	T3	19970301	ES 1992-917902	19920716
CA 2114796	C	20030513	CA 1992-2114796	19920716
PRIORITY APPLN. INFO.:			US 1991-750092	A 19910826
			WO 1992-US5976	W 19920716

AB The title optical fibers are prepared by coating with comps. containing fluorinated monoacrylates, fluorinated polyvinyl compds., adhesion-promoting monomers and optionally silane-free fluorinated acrylamides, non-fluorinated mono or polyvinyl compds., non-fluorinated silanes, photoinitiators, thermal stabilizers, and antioxidants. Thus, a composition containing (perfluorocyclohexyl)methyl acrylate, trimethoxysilylpropyl methacrylate, hexafluoropentamethylene diacrylate, stabilizers and an initiator gave a 25- μ m cladding on an optical fiber and showed optical loss 3.8 dB/km and numerical aperture 0.45 (at 633 nm for a 2-m fiber).

REFERENCE COUNT: 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 11 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN
 ACCESSION NUMBER: 1991:6078 CAPLUS
 DOCUMENT NUMBER: 114:6078
 ORIGINAL REFERENCE NO.: 114:1195a,1198a
 TITLE: Studies on the biosynthesis of clavulanic acid. Part 5. Absolute stereochemistry of proclavaminic acid, the monocyclic biosynthetic precursor of clavulanic acid
 AUTHOR(S): Baggaley, Keith H.; Elson, Stephen W.; Nicholson, Neville H.; Sime, John T.
 CORPORATE SOURCE: Chemother. Res. Cent., Beecham Pharm., Betchworth/Surrey, RH3 7AJ, UK
 SOURCE: Journal of the Chemical Society, Perkin Transactions 1: Organic and Bio-Organic Chemistry (1972-1999) (1990), (6), 1521-33
 CODEN: JCPRB4; ISSN: 0300-922X
 DOCUMENT TYPE: Journal
 LANGUAGE: English
 OTHER SOURCE(S): CASREACT 114:6078
 GI



AB Proclayaminic acid was synthesized by a route which indicated its constitution to be (2S,3R)-5-amino-3-hydroxy-2-(2-oxazolididin-1-yl)valeric acid (I). The spectroscopic properties of synthetic I were identical with those of natural I and both were converted into clavaminic acid by clavaminic acid synthase. An efficient synthesis of 3-hydroxyornithine derivs. was devised which allowed the separation of diastereoisomers and the resolution of a threo compound by the acylase [EC 3.5.1.11] from *Escherichia coli*. The β -lactam ring was subsequently elaborated by Michael addition of a protected 3-hydroxyornithine to acrylic acid followed by ring closure using PPh₃-di-2-pyridyl disulfide. Model reactions were carried out with enantiomerically pure threonine derivs. to confirm that the formation of the β -lactam moiety did not impair the integrity of the α - and β -chiral centers and that the enzymic deacylation reaction was capable of resolving the α -center of an α -amino- β -hydroxy acid. The enantiomeric purity of intermediates was determined using HPLC, ¹H NMR spectroscopy utilizing the chiral solvating reagents (R)- and (S)-1-(9-anthryl)-2,2,2-trifluoroethanol, and chiral GLC techniques.

L8 ANSWER 12 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:477514 CAPLUS

DOCUMENT NUMBER: 111:77514

ORIGINAL REFERENCE NO.: 111:13055a,13058a

TITLE: Purification of unsaturated carboxylic acid

isocyanatoalkyl esters by distillation

INVENTOR(S): Abe, Tetsuo; Yokoo, Hidejiro; Nozawa, Kaneo

PATENT ASSIGNEE(S): Showa Rodia Kagaku Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 6 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01042461	A	19890214	JP 1987-198155	19870810
JP 07049413	B	19950531		

PRIORITY APPLN. INFO.: JP 1987-198155 19870810

OTHER SOURCE(S): MARPAT 111:77514

AB The title esters, useful as monomers, are purified by distillation under continuous or intermittent addition of nitrite esters in the presence of Sn(2+) and/or Fe(2+) compds. to prevent popcorn polymerization CH₂:CMeCO₂H

(320

g) was gradually added to a solution of 300 g 2-oxazolidinone in toluene containing phenothiazine while bubbling with HCl at 60° over 60 min, and the reaction mixture was further bubbled with HCl at 60° for 30 min, and then heated at 80° while bubbling with COCl₂. After distilling off toluene, 230 g product containing CH₂:CMeCO₂CH₂CH₂NCO (I) was distilled with SnCl₂ and the HNO₂ ester (II) of HOCH₂CH₂OCH₂CH₂O₂Bu under dropwise addition of 50 g product containing II to give 115 g I. When the reaction product was distilled without addition of SnCl₂ and II, granules of polymerized matter were formed at the upper part of the distillation tower and polymer beads grew in the reaction mixture

L8 ANSWER 13 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:441476 CAPLUS
DOCUMENT NUMBER: 111:41476
ORIGINAL REFERENCE NO.: 111:7051a,7054a
TITLE: Silyl-containing vinyl resin compositions for coatings
INVENTOR(S): Yamashita, Takeshi; Konno, Ei-ju
PATENT ASSIGNEE(S): Dainippon Ink and Chemicals, Inc., Japan
SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.
CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 63275679	A	19881114	JP 1987-109898	19870507

PRIORITY APPLN. INFO.: JP 1987-109898 19870507

AB Storage-stable title compns. comprise polyoxazolines and vinyl polymers prepared from 1-100% vinyl monomers containing ≥ 1 unsatd. double bond and silyl ester group and 0-99% other vinyl monomers. Thus, styrene 200, Bu methacrylate 443, Acrister SL 150, and trimethylsilyl methacrylate 207 parts were polymerized in xylene in the presence of tert-Bu peroxyoctoate to give a polymer (I, number-average mol. weight 13,000) solution. A mixture of the solution

100, 1,3-bis(A2-oxazolin-2-yl)benzene 7.7, and HC(OMe)₃ 2.0 parts was diluted to give a storage-stable composition, which was sprayed on a Zn3(PO4)₂-treated steel plate and heated 30 min at 160° to form a coating with pencil hardness F-H, impact resistance 50, and good alkali, acid, and solvent resistance, vs., F-H, 20, poor, poor, and poor, resp., for a coating containing I (number-average mol. weight 850).

L8 ANSWER 14 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:173884 CAPLUS
DOCUMENT NUMBER: 110:173884
ORIGINAL REFERENCE NO.: 110:28867a,28870a
TITLE: End capping of growing species of poly(2-oxazoline) with carboxylic acid: a novel and convenient route to prepare vinyl- and carboxy-terminated macromonomers
AUTHOR(S): Miyamoto, Masatoshi; Naka, Kensuke; Tokumizu, Makoto; Saegusa, Takeo
CORPORATE SOURCE: Dep. Synth. Chem., Kyoto Univ., Kyoto, 606, Japan
SOURCE: Macromolecules (1989), 22(4), 1604-7
CODEN: MAMOBX; ISSN: 0024-9297
DOCUMENT TYPE: Journal
LANGUAGE: English

AB End-capping esterification of the growing oxazolinium end group in the polymerization of 2-oxazolines (2-Me and 2-Et derivs.) was accomplished by

the binary system of carboxylic acid and a proton scavenger of hindered base such as 2,6-lutidine. Reaction conditions for complete and quant. end capping were established. The methodol. was successfully applied to the synthesis of macromonomers having an acrylate end group (end capping with acrylic acid) and having a carboxylic acid end group (end capping with a dibasic acid). In addition, an A-B-A-type block copolymer between poly[(N-acylimino)ethylene] and poly(oxyethylene) was readily prepared by using poly(oxyethylene) dicarboxylic acid.

L8 ANSWER 15 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1988:168145 CAPLUS
DOCUMENT NUMBER: 108:168145
ORIGINAL REFERENCE NO.: 108:27663a,27666a
TITLE: Preparation of unsaturated fatty acid

2-isocyanatoethyl esters from 2-oxazolidinone,
unsaturated fatty acids, and phosgene
INVENTOR(S): Yokoo, Hidejiro
PATENT ASSIGNEE(S): Showa Rhodia Kagaku Co., Ltd., Japan
SOURCE: Jpn. Kokai Tokkyo Koho, 2
CODEN: JKXXAF
DOCUMENT TYPE: Patent
LANGUAGE: Japanese
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 63010750	A	19880118	JP 1986-152782	19860701
JP 07042263	B	19950510		
PRIORITY APPLN. INFO.:			JP 1986-152782	19860701

OTHER SOURCE(S): CASREACT 108:168145
AB Unsatd. lower fatty acid 2-isocyanatoethyl esters, useful as bifunctional monomers, were prepared by treating 2-oxazolidinone (I) with unsatd. fatty acids in the presence of HCl followed by treatment of the resulting product with COCl₂. Thus, 32 g CH₂:CMeCO₂H was added dropwise to a toluene solution containing 30 g I and phenothiazine under bubbling with HCl at 60° over 60 min, the reaction mixture was further stirred under HCl for 30 min, and then the resulting suspension was bubbled with COCl₂ at 80° till it became homogeneous solution to give 24 g CH₂:CMeCO₂CH₂CH₂NCO.

L8 ANSWER 16 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER: 1987:618053 CAPLUS
DOCUMENT NUMBER: 107:218053
ORIGINAL REFERENCE NO.: 107:35007a,35010a
TITLE: Fixation of lysine by its α-amino function to oligomethacrylic chains
AUTHOR(S): Laquerre, A.; Rabadeux, J. C.; Gueniffey, H.; Bruneau, C. M.
CORPORATE SOURCE: Fac. Sci., Univ. Maine Route, Laval, 72017, Fr.
SOURCE: European Polymer Journal (1987), 23(2), 113-16
CODEN: EUPJAG; ISSN: 0014-3057
DOCUMENT TYPE: Journal
LANGUAGE: French
AB The title fixation was achieved via treatment of oligomethacrylate chloroformate with H-Lys(CO₂CH₂Ph)-OMe.

L8 ANSWER 17 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER: 1985:221648 CAPLUS
DOCUMENT NUMBER: 102:221648
ORIGINAL REFERENCE NO.: 102:34793a,34796a
TITLE: Vinyl ester resins containing triazine or both triazine and oxazoline groups
INVENTOR(S): Hefner, Robert E., Jr.
PATENT ASSIGNEE(S): Dow Chemical Co., USA
SOURCE: U.S., 12 pp.
CODEN: USXXAM
DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 4515934	A	19850507	US 1984-590819	19840319
WO 8504176	A1	19850926	WO 1984-US1725	19841025
W: AU, BR, DK, FI, JP, KR, NO				

AU 8435567	A	19851011	AU 1984-35567	19841025
AU 555851	B2	19861009		
BR 8407303	A	19860325	BR 1984-7303	19841025
JP 61501396	T	19860710	JP 1984-504017	19841025
JP 03057930	B	19910903		
CA 1235845	A1	19880426	CA 1984-466386	19841026
EP 156959	A1	19851009	EP 1984-113039	19841030
EP 156959	B1	19890125		
R: AT, BE, CH, DE, FR, GB, IT, LI, NL, SE				
ZA 8408471	A	19860625	ZA 1984-8471	19841030
IL 73357	A	19880531	IL 1984-73357	19841030
AT 40391	T	19890215	AT 1984-113039	19841030
FI 8504184	A	19851025	FI 1985-4184	19851025
FI 81108	B	19900531		
FI 81108	C	19900910		
NO 8504596	A	19851118	NO 1985-4596	19851118
NO 162245	B	19890821		
NO 162245	C	19891129		
DK 8505339	A	19851119	DK 1985-5339	19851119
PRIORITY APPLN. INFO.:				
			US 1984-590819	A 19840319
			WO 1984-US1725	A 19841025
			EP 1984-113039	A 19841030

AB Heat-resistant vinyl ester resins are manufacture by esterification of aromatic epoxy resins containing oxazoline and (or) triazine groups with monounsaturated monoacids. Thus, 75.9 g Et3N was added in 20 min to 87.39 g CNBr and 342.45 g bisphenol A (I) in 950 mL Me2CO at -5 to -10°, and the reaction mixture was held 30 mins at -3 to +5° to give 337 g product mixture containing I 59.53, I monocyanate 35.01, and I dicyanate 5.46%

which was trimerized (335 g) 1.25 h at 177° in presence of 0.34 g 6% Co naphthenate to give a transparent, triazine ring-containing oligomer. This oligomer (250 g) was epoxidized with 763.05 g epichlorohydrin in 410.87 g iso-PrOH and 66.35 g water at 50-60° in presence of 171.53 g NaOH (added as a solution in 686.11 g water in 2 steps) to give a resin with epoxide equivalent weight 193.43. This resin (300 g) was esterified with 129.38 g methacrylic acid in presence of 0.172 g hydroquinone and 0.375 g CrCl3 catalyst (added as a 33.33% aqueous solution) at 90-117° to give a vinyl ester resin containing 1.298% carboxylic acid, which was mixed with 0.172 g phenothiazine, 240.63 g styrene, 1% Bz2O2, and 0.05% PhNMe2, cured at room temperature 24 h after cure exotherm subsided, and post cured 2 h at 100° to give a cast molding with heat-distortion temperature 105.5° and tensile strength 64.1 kPa, compared with 101.1 and 62.1, resp., for a molding prepared from a methacrylated mixture of epoxy novolak resin and I diglycidyl ether.

REFERENCE COUNT: 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L8 ANSWER 18 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1981:480671 CAPLUS

DOCUMENT NUMBER: 95:80671

ORIGINAL REFERENCE NO.: 95:13639a,13642a

TITLE: Total synthesis of the antibiotic sparsomycin, a

modified uracil amino acid monoxodithioacetal

AUTHOR(S): Ottenheijm, Harry C. J.; Liskamp, Rob M. J.; Van

Nispen, Simon P. J. M.; Boots, Hans A.; Tijhuis,

Marian W.

CORPORATE SOURCE: Dep. Org. Chem., Univ. Nijmegen, Nijmegen, Neth.

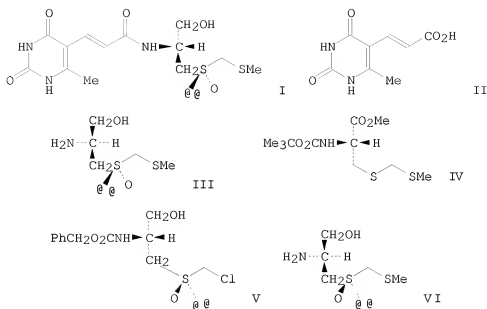
SOURCE: Journal of Organic Chemistry (1981), 46(16), 3273-83

CODEN: JOCEAH; ISSN: 0022-3263

DOCUMENT TYPE: Journal

LANGUAGE: English

GI



AB The total syntheses of sparsomycin (I), a naturally occurring antibiotic and antitumor substance, and its three stereoisomers are described for the first time. In a convergent approach, the carboxylic acid II and the amine III were synthesized followed by amide formation. The acid II was prepared (23% yield) from 6-methyluracil by coupling of the 4-formyl derivative with $\text{Ph}_3\text{P:CHCO}_2\text{Et}$. The synthesis of III was accomplished by the reaction of a cysteine α -halosulfoxide derivative with MeSNa . Alternatively, oxidation of the dithio acetals, e.g. IV, was unsatisfactory, yielding predominantly the undesired regioisomers. Procedures are given for the preparation and separation of the α -halosulfoxide diastereomers e.g. V. By use of these procedures, the amino alc. monoxodithioacetals III and VI were prepared in five steps (40% yield) from a D-cystine derivative having the SC chirality of sparsomycin. Finally, I and the SC diastereomer were prepared (40% yield) by mixed anhydride coupling of II with III and VI, resp. In addition, syntheses of the RC enantiomer and the corresponding diastereomer are described. The CD spectra of I and its three stereoisomers are also discussed.

L8 ANSWER 19 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1979:457006 CAPLUS
 DOCUMENT NUMBER: 91:57006
 ORIGINAL REFERENCE NO.: 91:9239a,9242a
 TITLE: 2-Guanidino-4(5H)-imidazolone derivatives
 INVENTOR(S): Honna, Ryuji; Ogawa, Kazuo; Toratani, Keiko;
 Hashimoto, Sadao; Suzue, Takashi
 PATENT ASSIGNEE(S): Taiho Yakuhiin Kogyo K. K., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

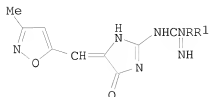
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 5401997	A	19790215	JP 1977-83207	19770711

JP 62024434
PRIORITY APPLN. INFO.:
GI

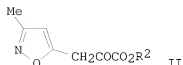
B 19870528

JP 1977-83207

A 19770711



I



II

AB The title derivs. I (R, R1 = H, H; Me, Me; Bu, H; PhCH2CH2, H) were prepared by reaction of II (R2 = alkyl) with H2NC(:NH)NHC(:NH)NRR1. Thus, a mixture of 3-methylisoxazole-5-carboxaldehyde 5.6, BzNHCH2CO2H 9, Ac2O 17.4, and NaOAc 4.1 g was stirred 0.5 h at 80-90° to give 55% α -benzoylamino- β -(3-methyl-5-isoxazolyl) acrylic acid, which was refluxed with HCl-saturated MeOH 2 h to give 70% II (R2 = Me; III). A mixture of 0.46 g Na and 3.36 g H2NC(:NH)NHC(:NH)NH2.H2SO4.2H2O in MeOH was kept 1 h at room temperature, refluxed 1 h, 3.66 g III added, and the whole refluxed 3 h to give 43% I (R = R1 = H; IV). Hypoglycemic and blood free fatty acid level lowering data of IV were given in rats.

L8 ANSWER 20 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1974:570560 CAPLUS

DOCUMENT NUMBER: 81:170560

ORIGINAL REFERENCE NO.: 81:26395a,26398a

TITLE: Resin for curing by exposing to ionizing radiation a mixture of a vinyl ester resin, an alkenyl aromatic monomer, and a 2-oxazoline or guanidine

INVENTOR(S): Mani, Inder

PATENT ASSIGNEE(S): Dow Chemical Co.

SOURCE: U.S., 3 pp.

CODEN: USXXAM

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 2

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 3810825	A	19740514	US 1971-143266	19710513
US 3882004	A	19750506	US 1973-391112	19730827

PRIORITY APPLN. INFO.: US 1971-143266 A3 19710513

AB Addition of .geq.0.3% N-containing curing accelerator to a thermosettable vinyl aromatic monomer-vinyl ester resin prepolymer mixture reduces the dosage of ionizing radiation needed for curing. The accelerators include 2-oxazolines, guanidines and alkylamines. Maleic anhydride [108-31-6] treated with 2-hydroxypropyl acrylate [999-61-1] gave the acid ester which was treated with 1,4-butanediol diglycidyl ether, mixed with styrene, and irradiated. The resin [53108-91-1] system required 2.8-3.0 Mrad dose of electron beam irradiation while the dose was 1.3-1.4 Mrad for a similar one containing 3% tetramethylguanidine [80-70-6].

L8 ANSWER 21 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1974:97478 CAPLUS

DOCUMENT NUMBER: 80:97478

ORIGINAL REFERENCE NO.: 80:15685a,15688a

TITLE: Unsaturated ester of polyhydroxy-bis-oxazolines

INVENTOR(S): Mileo, Jean C.; Sillion, Bernard

PATENT ASSIGNEE(S): Commissariat a l'Energie Atomique; Institut Francais du Petrole, des Carburants et Lubrifiants
 SOURCE: Ger. Offen., 20 pp.
 CODEN: GWXXBX
 DOCUMENT TYPE: Patent
 LANGUAGE: German
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 2311387	A1	19730927	DE 1973-2311387	19730308
FR 2176192	A5	19731026	FR 1972-8730	19720313
BE 796433	A1	19730910	BE 1973-1004867	19730308
NL 7303459	A	19730917	NL 1973-3459	19730312
IT 981298	B	19741010	IT 1973-21455	19730312
JP 49000263	A	19740105	JP 1973-29368	19730313
GB 1385842	A	19750305	GB 1973-11925	19730313
			FR 1972-8730	A 19720313

PRIORITY APPLN. INFO.:

AB Polymethylenebis[(hydroxymethyl)oxazoline] acrylates were prepared and used in electron beam-curable coating compns. Thus, azelaic acid [123-99-9] was condensed with 2-amino-2-methyl-1,3-propanediol [115-69-5] to give heptamethylenebis[4-(hydroxymethyl)-4-methyloxazoline] [38836-05-4], which was esterified with acrylic acid [79-10-7] to give heptamethylenebis[4-(hydroxymethyl)-4-methyloxazoline] diacrylate (I) [50979-14-1]. A solution of 7 g I and 3 g styrene was coated on stainless steel and irradiated with a 500 kV electron beam in N, giving a cured thickness of 120 μ and average hardness 5H at a dose of 10 Mrad.

L8 ANSWER 22 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1973:5539 CAPLUS
 DOCUMENT NUMBER: 78:5539
 ORIGINAL REFERENCE NO.: 78:901a,904a
 TITLE: Thermosetting acrylics containing oxazoline groups
 INVENTOR(S): Dowbenko, Rostyslaw
 PATENT ASSIGNEE(S): PPG Industries, Inc.
 SOURCE: U.S., 6 pp. Division of U.S. 3,609,161 (CA 76:26516s).
 CODEN: USXXAM
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 3692757	A	19720919	US 1970-89116	19701112
			US 1970-89116	A 19701112

PRIORITY APPLN. INFO.:

AB Unsatd., polymerizable oxazolines were prepared by dehydrating a hydroxyalkyl aminoethanol, and esterifying the alc. with a monocarboxylic acid, ester, or acid halide. The oxazolines were polymerized or copolymd. to form thermosetting films, useful as solvent-resistant protective coatings. Thus, a mixture of 68 parts AcOH and 105.1 parts 2-methyl-2-amino-1,3-propanediol was refluxed 4 hr at 158-203.deg. to form 168.4 parts 2,4-dimethyl-4-hydroxymethyl-2-oxazoline, which was treated with methacrylic acid in benzene to obtain 2,4-dimethyl-2-oxazoline-4-ylmethyl methacrylate (I) [35061-72-4]. I homopolymd. or copolymd. in xylene and BuOH with Me methacrylate, 2-hydroxyethyl methacrylate, acrylic acid, and 2-ethylhexyl acrylate in the presence of azobisisobutyronitrile to yield flexible, xylene-resistant coating compns.

L8 ANSWER 23 OF 25 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1959:111733 CAPLUS

DOCUMENT NUMBER: 53:111733
ORIGINAL REFERENCE NO.: 53:20021e-i, 20022a-i, 20023a-i, 20024a-i, 20025a-i
TITLE: Griseofulvin. XVII. Synthesis of
7-chloro-4,6-dimethoxy-2'-methylgrisan-3,4'-dione
AUTHOR(S): MacMillan, J.
CORPORATE SOURCE: Imp. Chem. Inds. Ltd., Welwyn, UK
SOURCE: Journal of the Chemical Society (1959) 2211-23
CODEN: JCSOA9; ISSN: 0368-1769

DOCUMENT TYPE: Journal
LANGUAGE: Unavailable

GI For diagram(s), see printed CA Issue.

AB The synthesis is described of grisan and 3-coumaranone spirocyclopentane
analogs of some griseofulvin (I) derivs. and of racemates of some I
degradation products. (All m.ps. are corrected; absorption spectra determined

in

EtOH.) Et 6-hydroxy-3-methylcoumarone-2-carboxylate (II), m. 179°
(EtOH), ν 1700 cm.⁻¹, λ 313, 282, 241, 210 μ (log ϵ
4.30, 4.07, 3.90, 4.24), was prepared in 60% yield by the method of Hantzsch
[Ber. 19, 292(1886)]. Treatment of II with Me₂SO₄ gave 30%
6-methoxy-3-methylcoumarone-2-carboxylic acid (III), m. 184°, and
25% III Et ester (IV), m. 74-5°, ν 1700 cm.⁻¹, λ 312,
285, 243 μ (log ϵ 4.44, 4.13, 3.97); esterification
of III gave IV. 6-Methoxy-3-methylcoumarone-2-acetic acid (V) chloride
and AcCH(MgOEt)CO₂Et (VI) gave Et α -acetyl- β -oxo-6-methoxy-3-
methylcoumarone-2- γ -butyrate (VII), oil, ν 1714, 1623, and 1587
cm.⁻¹; Cu derivative (VIII), m. 175° (MeOH), ν 1715, 1692, 1671,
1587 cm.⁻¹ VII (7 g.) and 15 ml. concentrated H₂SO₄ kept 3 days at -5°,
diluted with 50 ml. H₂O, extracted with Et₂O, the Et₂O solution extracted with

aqueous

NaHCO₃, and the alkaline extract acidified gave 200 mg. V, m. 139-40°;
recovery from the Et₂O solution gave a pale yellow oil (IX). IX in EtOH
passed through a C column, the resulting material triturated for a long
time under petr. ether (X) at -50°, and the solid further
recrystd. at room temperature gave 2.8 g. Et
6-methoxy-3-methylenecoumaran-2-spiro-1'-(2'-methyl-4'-oxocyclopent-2'-en-
3'-carboxylate) (XI), m. 122°, ν 1745, 1707, 1649, 1615, 1595
cm.⁻¹, λ 327, 263, 231, 214 μ (log ϵ 3.98, 4.12, 4.35,
4.31); 2,4-dinitrophenylhydrazones, m. 175-6° (C₆H₆-X). Cyclization
of VIII with H₂SO₄ gave 45% XI. XI (150 mg.) hydrogenated at room temperature
and pressure with a catalyst prepared in situ from 7 mg. PdCl₂ and 75 mg. C
in 10 ml. AcOH (1 mole H absorbed), the recovered gum chromatographed in
Et₂O on Al₂O₃, eluted with Et₂O, and the product repeatedly triturated
with X at 0° gave 105 mg. corresponding 3-Me analog, m.
77-8°, ν 1738, 1698, 1610 cm.⁻¹, λ 320, 302, 281,
.apprx. 263, 228 μ (log ϵ 2.78, 3.45, 3.59, 3.72, 4.25);
2,4-dinitrophenylhydrazones, m. 160° (C₆H₆-petr. ether). A stream
of O₃-O passed into 500 mg. XI in 30 ml. CCl₄ until the solution became
opaque, the filtered solution evaporated, the residual gum mixed with 25 ml.

H₂O,

kept 24 hrs., and the aqueous phase distilled into saturated dimedon (XII)
solution in

10% aqueous alc. gave XII derivative of CH₂O, m. 191-2°; an Et₂O solution of
the residual gum chromatographed on Al₂O₃, the lowest bright-blue
fluorescent band eluted with Et₂O and the recovered product boiled with X
(110 mg. solid obtained); the solid combined with 100 mg. slightly impure
material from the following band fluorescing greenish-blue and recrystd.
(petr. ether) gave 200 mg. corresponding 3-oxo compound (XIII), m.
115-17°, ν 1751, 1715, and 1701 cm.⁻¹, λ 321, 275, 232
 μ (log ϵ 3.98, 4.21, 4.23); mono-2,4-dinitrophenylhydrazones,
m. 243° (decomposition) (C₆H₆-petr. ether). Oxidation of 100 mg. XIII with
Fehling's solution gave 25 mg. 2,4-HO(MeO)C₆H₃CO₂H, m. 154-5°. XIII
(70 mg.) and 17.5 ml. 2N HCl and 14 ml. EtOH refluxed 6 hrs. under N
cooled (15 mg. solid separated), EtOH removed in vacuo (15 mg. solid and 30

mg. XIII separated), the combined product purified through the Na salt, and recrystd. (EtOH) gave 23 mg. corresponding acid (XIV), m. 175-7°, ν 1716 and 1697 cm.⁻¹, λ 318, 273, 232, 210 m μ (log ϵ 4.05, 4.25, 4.30, 4.45). XIV (70 mg.) heated 10 min. at 190-200° in an N atmosphere, the residue sublimed in situ in vacuo, and the sublimate recrystd. (aqueous Me₂CO) gave 40 mg. 6-methoxy-3-oxocoumaran-2-spiro-1'-(2'-methylcyclopent-2'-en-4'-one) (XV), m. 193-5°, ν 1713 cm.⁻¹, λ 320, 273, 232, 209 m μ (log ϵ 3.92, 4.07, 4.16, 4.40); mono-2,4-dinitrophenylhydrazone, m. 238-40° (EtOAc). XV (15 mg.) in 1.5 ml. EtOAc added to a previously catalyst prepared from 10 mg. PdCl₂ and 40 mg. C in 1 ml H₂O, the mixture shaken with H at room temperature and pressure until absorption ceased (2.4 moles H absorbed in 5 min.), and the product (XVI) (13 mg.) recovered; XVI (13 mg.) chromatographed in C₆H₆ on Al₂O₃, a bright-blue fluorescent band eluted with C₆H₆, the product (2 mg.) recovered, sublimed, and crystallized (petr. ether) gave 0.8 mg. 6-methoxy-3-oxocoumaran-2-spiro-1'-(2'-methylcyclopentane), m. 167-8°, identical with the product from condensation of 6-methoxycoumaran-3-one and CH₂Br(CH₂)₂CHBrMe (cf. preceding abstract). β -Propiolactone (XVII) (8.64 g.) and 19.44 g. 6-methoxy-3-methylcoumarone refluxed 6 hrs. at 200°, the cooled residue ground with aqueous Na₂CO₃, the mixture filtered, the Et₂O-washed filtrate acidified, the precipitate filtered off, sublimed in vacuo, and the sublimate crystallized (dilute EtOH) gave 4.9 g. 6-methoxy-3-methyl-coumarone-2- β -propionic acid (XVIII), m. 137°, ν 1707 cm.⁻¹, λ 292 and 253 m μ (log ϵ 3.80, 4.16). 6-Methoxy-3-methylcoumarone-2-acetic acid (440 mg.) and 1 ml. SOCl₂ was warmed gently until a clear solution was obtained, volatiles removed in vacuo below 30°, the residual oil in 10 ml. Et₂O added at 0° to a 3-fold excess CH₂N₂ in 15 ml. Et₂O, the solution kept 24 hrs. at 0°, and the Et₂O removed in vacuo; the residual crude diazo ketone (m. 118-23°) in 10 ml. dioxane added dropwise with stirring to 2 g. Ag₂O in 1 g. Na₂S₂O₃ and 1 g. K₂CO₃ in 20 ml. H₂O at 50-60° (N evolved), after 20 min. the mixture heated 10 min. at 70° and 5 min. at 90°, filtered, the filtrate acidified with dilute HNO₃, the oil triturated with H₂O, the resulting semisolid sublimed in vacuo, and the sublimate crystallized (dilute EtOH) gave 60 mg. XVIII, m. 137-8°. XVIII treated with PCl₅ gave the acid chloride (XIX), green oil; amide, plates (C₆H₆-petr. ether). XIX in 500 ml. Et₂O mixed with a suspension of VI [from 2.7 g. AcCH₂CO₂Et (XX), 1.2 ml. EtOH, and 0.48 g. Mg in 50 ml. Et₂O], the mixture refluxed 24 hrs., excess dilute AcOH added, the separated

Et₂O

layer washed with aqueous NaHCO₃ (from the alkaline wash 200 mg. XVIII was isolated on acidification), concentrated, the oil (4.2 g.) chromatographed in C₆H₆ on Al₂O₃ and eluted with C₆H₆ gave, from a narrow fluorescent band, 90 mg. oil, which, on crystallization (petr. ether), afforded 60 mg. XVIII Et ester (XXI), m. 51-2°, ν 1737, 1624, 1594 cm.⁻¹, identical with material obtained by esterifying XVIII, and, from a greenish-red band, 3.1 g. oil, which, heated 4 hrs. at 100-10°/10-4 mm., yield 2.2 g. crude oxo ester (XXII), giving an intense Fe+++ reaction in EtOH and forming a Cu chelate. Crude XXII (12 g.) in 25 ml. concentrated H₂SO₄ kept 8 days at 0°, worked up as described above for the lower homolog, and the alkaline extract acidified gave a yellow precipitate which decomposed on filtration;

the unfiltered mixture extracted with C₆H₆, the extract passed through silica, and

eluted with C₆H₆ gave 600 mg. XVIII, m. 133-5°; the Et₂O solution concentrated, the oil (4.5 g.) chromatographed in Et₂O on Al₂O₃, and eluted

with

Et₂O gave, from a narrow band, 20 mg. XXI, and, from a pale blue band, 4.1 g. oil, which, crystallized (EtOH) and recrystd. (aqueous EtOH), afforded 3.8 g. Et

6-methoxy-2'-methyl-3-methylene-4'-oxogris-2'-ene-3'-carboxylate (XXIII), m. 97-8°, ν 1728, 1668, 1641, 1614 cm.⁻¹, λ 326, 316,

273, 260, 233, 215 μ (log ϵ 4.05, 4.09, 4.18, 4.43, 4.37); 2,4-dinitrophenylhydrazones, m. 182° (decomposition). Crude XXII (1.5 g.) mixed with 20 g. polyphosphoric acid (XXIV), kept 4 days at room temperature, and worked up gave 700 mg. XXIII, 30 mg. XXI, and no acidic product. XXIII (1.0 g.) in 30 ml. CCl_4 ozonized as above (CH_2O isolated with XII), the product chromatographed in Et₂O on Al₂O₃, the yellow gum (510 mg.) treated with EtOH, seeded, and the solid recrystd. (dilute EtOH) gave Et 6-methoxy-2'-methyl-3,4'-dioxogris-2'-ene-3'-carboxylate (XXV), m. 110-11°, ν 1735, 1704, 1686 cm^{-1} , λ 320, 272, 231, 208 μ (log ϵ 3.95, 4.12, 4.17, 4.42), giving an intractable precipitate with Brady reagent. XXV (3 g.) refluxed 4 hrs. in an N atmosphere with 300 ml. EtOH and 350 ml. 2N HCl (0.6 mole CO_2 evolved), the solution concentrated in vacuo, the aqueous residue extracted with Et₂O, the extract washed with aqueous NaHCO₃ (recovery gave only a trace of acidic gum), concentrated, the residual oil (1.6 g.) chromatographed in Et₂O on Al₂O₃, and eluted with Et₂O gave, from a blue band, 1.5 g. 6-methoxy-2'-methylgris-2'-ene-3,4'-dione, gum, ν 1713 and 1667 cm^{-1} , λ , 318, 272, 232, 209 μ (log ϵ 3.98, 4.13, 4.22, 4.45) [2,4-dinitrophenylhydrazones, m. 230° (EtOAc)], and, from a dark blue band, 20 mg. XXV. 3,2,4,6-Cl(HO)(MeO)2C₆H₄C (XXVa) (46.0 g.), 66.4 g. $\text{CH}_2\text{BrCO}_2\text{Et}$, and 56.0 g. anhydrous K_2CO_3 in 700 ml. Me₂CO refluxed 72 hrs., filtered, the filtrate evaporated in vacuo, the oil dissolved in 200 ml. EtOH, the solution treated with Et₂O, and the precipitate recrystd. (EtOH) gave 56.6 g. 2,6,3,5-Ac(Cl)-(MeO)2C₆H₃CH₂CO₂Et (XXVI), m. 72°, ν 1763, 1735, 1684 cm^{-1} , λ 305, 292, 258, 223 μ (log ϵ 3.35, 3.43, 3.70, 4.21); 2,4-dinitrophenylhydrazones m. 136° (EtOAc). With $\text{CH}_2\text{-BrCO}_2\text{Me}$ in the above experiment, there was obtained the corresponding Me ester (XXVII), m. 103-4° (MeOH), ν 1760 and 1680 cm^{-1} , λ 305, 293, 268, 233 μ (log ϵ 3.35, 3.42, 3.66, 4.20). XXVI (47.4 g.) in 200 ml. EtOH, 300 ml. H₂O, and 14 ml. H₂SO₄ refluxed 4 hrs., diluted with H₂O until cloudy, kept overnight at 5%, the precipitate filtered off, washed with aqueous Na₂CO₃ (19.7 g. XXVI remained), and the alkaline extract acidified gave 19.1 g. acid (XXVIII), m. 144° (C₅H₆ or H₂O), ν 1751 and 1659 cm^{-1} , λ 301, 292, 258, 224 μ (log ϵ 3.42, 3.39, 3.66, 4.19). XXVII (500 mg.) and 20 ml. 3N HCl refluxed 6.5 hrs., the solid filtered off, dissolved in warm aqueous Na₂CO₃, the solution treated with C, filtered, and the filtrate acidified at 0° gave 311 mg. XXVIII cyclic form (XXIX) (R = H) (XXX), m. 143-4° (dilute EtOH), ν 1746 cm^{-1} , λ 275 and 240 μ (log ϵ 3.09, 3.95); mixts. of XXVIII and XXX m. 115-40°; a 1:1 mixture (XXXI), m. 119-20°, behaved as a pure compound (see below). XXVII (1.5 g.) and 75 ml. N HCl refluxed 4 hrs. and worked up gave 1.1 g. XXX, m. 119-20° (C₆H₆), ν 1749 and 1659 cm^{-1} , identical with a 1:1 mixture XXVIII and XXX. XXX (29 mg.) in 5 ml. Et₂O treated overnight with 1 mole CH_2N_2 in 5 ml. Et₂O and worked up gave 24 mg. XXIX (R = Me) (XXXII), m. 97° (MeOH), ν 1742 cm^{-1} , λ .apprx.275 and 235 μ (log ϵ 3.07, 3.98). XXVIII treated similarly gave the isomeric open-chain acid (XXXIII), m. 103-4°, mixed m.p. with XXXII 78-85°. XXXIII or XXXII (17.3 g.), 34 g. fused NaOAc, and 100 ml. Ac₂O heated under reflux until CO_2 evolution ceased (30 min.), the cooled solution diluted with H₂O, made alkaline with Na₂CO₃, the product filtered off, passed through a column of Al₂O₃ in 3 l. C₆H₆ recovered, and crystallized (EtOH) gave 14 g. 7-chloro-4,6-dimethoxy-3-methylcoumarone (XXXIV), m. 148°, ν 1626, 1611, 1591 cm^{-1} , λ 320, .apprx.275, 263, 220, 216 μ (log ϵ 2.42, 4.00, 4.19, 4.56, 4.56). XXVI (15.8 g.) refluxed 1 hr. with 1 g. Na in 50 ml. EtOH, the cooled solution diluted with 750 ml. H₂O, the precipitate (2.7 g.) filtered off, and crystallized (EtOH) gave Et 7-chloro-4,6-dimethoxy-3-methylcoumarone-2-carboxylate (XXXV), m. 172-3°, ν 1712 cm^{-1} , λ 312, 298, 245, 230 μ (log ϵ 4.32, 4.37, 4.34, 4.30); acidification of the filtrate with dilute

HCl gave 11.2 g. XXXVIII, m. 143°. XXXV (895 mg.) warmed 10 min. at 40° with 6 g. NaOH in 100 ml. 50% aqueous EtOH, diluted with ice and H₂O, the solution acidified, and the precipitate (790 mg.) sublimed in vacuo gave

the

corresponding acid (XXXVI), m. 310° (decomposition), ν , 1671 cm.⁻¹, λ 293 and 238 μ (log ϵ 4.26, 4.31); Me ester, m. 196° (MeOH). XXXVI converted to the acid chloride (SOCl₂ plus C₅H₅N) and the latter treated with CH₂N₂-Et₂O gave the diazo ketone (XXXVII), m. 118-20° (Et₂O). Attempted Wolff rearrangement of 600 mg. XXXVII as above gave 530 mg. XXXVI, m. 250-60° (decomposition). XXXVI (200 mg.) in 0.5 ml. quinoline heated 5 min. at 190-5° with 100 mg. Cu bronze and worked up gave 150 mg. XXXIV, m. 143°. Anhydrous HCN (200 ml.) added to 142 g. XXXIV in 6 l. Et₂O at 0°, a fast stream of dry HCl passed in at 0-5° until absorption ceased (6 hrs.), the solution kept overnight at 0°, the aldimine-HCl (XXXVIII) filtered off, washed with Et₂O, heated with 10 l. H₂O at 100°, and the product (XXXIX) (110 g.) filtered off (after 24 hrs. a 2nd crop XXXVIII separated from the Et₂O mother liquor which yielded 27 g. addnl. XXXIX on hydrolysis); XXXIX in C₆H₆ passed through Al₂O₃, crystallized (C₆H₆, EtOH, or Me₂CO), sublimed, and the sublimate recrystd. (Me₂CO) gave the 2-OHC derivative (XL) m. 183°, μ 1662 cm.⁻¹, λ 341, 312, 253 μ (log ϵ 4.24, 4.16, 4.28) [2,4-dinitrophenylhydrazones m. 312° (Me₂CO); azlactone m. 235° (EtOAc), giving on alkaline hydrolysis 7-chloro-4,6-dimethoxy-2,3-dimethylcoumarone (XLI), m. 290° (MOH), μ and 1596 cm.⁻¹]. XL oxidized with KMnO₄ in aqueous Me₂CO gave XXXVI, m. 310° (decomposition). XL (2.26 g.) and 720 mg. XVII refluxed 4 hrs., cooled, the melt dissolved in 200 ml. Et₂O, the solution filtered, the filtrate extracted with aqueous Na₂CO₃, and the

alkaline extract

acidified gave 720 mg. 7-chloro-4,6-dimethoxy-3-methylcoumarone-2- β -propionic acid (XLII), m. 123-30° (chromatography followed by recovery from the Ag salt); a sample of XLII was converted by Et₂O-CH₂N₂ to the Me ester (XLIII), m. 102° (MeOH), ν 1739 cm.⁻¹, λ 320, 294, 278, 265, 226, 223 μ (log ϵ 2.61, 3.30, 4.05, 4.27, 4.53, 4.52, 4.52). The remaining XLII (500 mg.) methylated as above, the product chromatographed in C₆H₆ on Al₂O₃, a pale-blue band eluted with Et₂O, the recovered ester (350 mg.) hydrolyzed with N alc. HCl, the compound sublimed in vacuo, and recrystd. (aqueous EtOH) gave XLII, m. 134-5°, ν (Nujol mull) 1722 and 1706 cm.⁻¹, (in CHCl₃) 1716 cm.⁻¹ XXXIX (4 g.), 6.7 g. CH₂(CO₂H)₂, 330 mg. piperidine, and 17 ml. C₅H₅N heated 1 hr. at 100° and then 10 min. at 120°, the cooled solution acidified, the precipitate (6.1 g.) extracted with aqueous Na₂CO₃ (1.3 g. XXXIV remained), and the alkaline extract acidified gave 4.75 g. 7-chloro-4,6-dimethoxy-3-methylcoumarone-2-acrylic acid (XLIV), m. 252° (C₆H₆), ν 1692 and 1676 cm.⁻¹, λ 329 and 255 μ (log ϵ 4.44, 4.04); Me ester, m. 179° (MeOH), ν 1709 cm.⁻¹ (a) XLIV (102 mg.) in 50 ml. AcOH added to 0.2 g. Raney Ni in 2 ml. EtOH, hydrogenated at room temperature and pressure until absorption ceased (1 mole H absorbed in 8 hrs.), the mixture filtered, the filtrate concentrated to 5 ml. in vacuo, 100 ml. H₂O added, stored overnight at 5°, the solid (60 mg.) collected, stirred 1 hr. with cold dilute HCl, the mixture filtered, the precipitate (35 mg.) purified through the Na salt,

and

crystallized (petr. ether) gave XLII, identical with XLII as prepared above.

(b)

Crude XLIV (93 g.) in 3.5 l. 0.1N NaOH hydrogenated at room temperature and pressure in the presence of 50 g. 5% Pd-SrCO₃ (absorption ceased after 8 hrs. when 0.64 mole H was absorbed), 10 g. C added, the filtered solution acidified (38 g. solid obtained and an addnl. 5 g. by repptg. the sticky material from aqueous NaOH), and the product chromatographed in C₆H₆ on silica gave 39 g. XLII, m. 133-5°, identical with XLII prepared in a. (c) XLIV (56 g., m. 252°) in 2.3 l. 0.1N NaOH hydrogenated as in b (2 moles H absorbed in 1 hr.), the acidified solution extracted with Et₂O, the

exts.

concentrated, the combined solids therefrom treated with CH₂N₂-Et₂O, the resulting oil (50.7 g.) chromatographed in C₆H₆ on Al₂O₃, and a single bright blue band eluted with C₆H₆ gave 6.3 g. product (XLV), needles, m. 104-6° (C₆H₆-petr. ether), and 3.6 g. product (XLVI), prisms, m. 98-100° (MeOH), and, on further elution, 37 g. gum (XLVII). XLV crystallized (MeOH) gave Me 7-chloro-4,6-dimethoxy-3-methylcoumaran-2-β-propionate, prisms, m. 107-8°, ν 1731 cm.⁻¹, hydrolyzed with boiling N NaOH to the corresponding acid, m. 147-8°, ν 1716 cm.⁻¹ XLVI recrystd. (MeOH) gave 2.7 g. XLIII, m. 102°, alkaline hydrolysis giving 1.9 g. XLII. XLVII rechromatographed, the single band eluted in 10 equal fractions with C₆H₆, and fractions 2 and 8 rechromatographed; fraction 2 gave essentially Me 4,6-dimethoxy-3-methylcoumaran-2-β-propionate (XLVIII), uncrystallizable gum, ν 1728 cm.⁻¹, alkaline hydrolysis yielding the acid (XLIX), m. 114° (1:1 Et₂O-petr. ether); fraction 8 hydrolyzed with NaOH, the product crystallized (1:1 C₆H₆-petr. ether), the crystals hand-picked, and recrystd. gave 4,6-dimethoxy-3-methylcoumarone-2-β-propionic acid (L), m. 124°. XLIX (2 g.) and 200 mg. S heated 20 min. at 220° and 5 min. at 280°, the product continuously extracted with X, and the resulting crystallized extract hand-picked gave 320

mg.

recovered XLIX and 540 mg. L, m. 124°. XLVIII (157 mg.) and 135 mg. o-chloranil in 5 ml. CHCl₃ stored 1.5 hrs. in the dark, the recovered product chromatographed in C₆H₆ on Al₂O₃, the pale blue band eluted with C₆H₆, and the product (135 mg.) crystallized (1:1 Et₂O-petr. ether and then MeOH) gave the coumarone-tetrachloropyrocatechol complex (LI), m. 117°. LI hydrolyzed by alkali and the product crystallized (1:1 Et₂Ox) gave L, m. 122-4°. XLII (23.5 g.) in 500 ml. dry CHCl₃ treated with 20 g. PCl₅, the mixture refluxed 0.5 hr., volatile products removed at 30° in vacuo, and the residue (LII) kept 4 hrs. at 12 mm., LII dissolved in 1.5 l. Et₂O, 12 g. VI added, the mixture refluxed 24 hrs., 250 ml. 40% AcOH added, the Et₂O layer separated, extracted with Na₂CO₃ solution

(the

extract afforded 5.9 g. XLII), and concentrated; the residual oil (25 g.) chromatographed in Et₂O on Al₂O₃ and eluted gave, from a bright-blue band, 800 mg. XLII Et ester (LIII), m. 74-6° (EtOH), and, from an orange-red band, 18 g. orange oil (LIV).

LIV heated 2 hrs. at 110°/10-4 mm. (excess XX removed) gave 14 g. gum (LV), giving a red color with alc. FeCl₃ and forming no stable ketonic derivs. LV (12 g.) in 100 ml. Et₂O shaken with saturated aqueous Cu(OAc)₂ deposited 2.8 g. blue crystals (LVI); the Et₂O layer evaporated and the residual gum fractionally crystallized gave 6 g. product (LVII) and 1.8 g. Et α-acetyl-β-oxo-7-chloro-4,6-dimethoxy-3-methylcoumarone-2-γ-carboxylate (LVIII) β-Cu chelate (LIX), m. 98-100° (EtOH), converted to the α-Cu isomer (LX) on continued heating at 120-30°; LVI and LVII combined and crystallized (CHCl₃) gave LX, m. 164°. (a) LIX (1.4 g.) and 20 g. XXIV kept 10 days at room temperature, diluted with 50 ml. ice and H₂O, extracted with Et₂O, the extract washed with

aqueous

Na₂CO₃, dried, evaporated, the residual gum (1.18 g.) chromatographed in C₆H₆ on Al₂O₃, and the column eluted with C₆H₆ gave (1) from a narrow blue band, 5 mg. LIII, m. 72-3°, (2) from a greenish-blue band, a gum, which, kept in EtOH, deposited 175 mg. product (LXI), m. 137° and (3) from a blue band, a gum, which, kept in X, deposited 278 mg. product (XLII), m. 129-31°. LXI and LXII combined, recrystd. (EtOH), the crystals (410 mg.) sublimed in vacuo, and the sublimate recrystd. (MeOH) gave Et 7-chloro-4,6-dimethoxy-2'-methyl-3-methylene-4'-oxogris-2'-ene-3'-carboxylate (LXIII), m. 137°, ν 1731, 1680, 1638, 1610 cm.⁻¹, λ₂₈₂, .apprx.275, and 253 μ (log ε 4.30, 4.28, 4.57).

(b) LX (500 mg.) and 20 g. XXIV heated 1 hr. at 70-80° and worked up as above gave 210 mg. LXIII, m. 137°. (c) LX (1.8 g.) and 30 g.

XXIV kept 10 days at room temperature, worked up as above, and the alkaline extract

acidified gave 600 mg. XLII; some Cu derivative (600 mg.) was regenerated by shaking the acid with aqueous Cu(OAc)₂; the remaining gum (320 mg.) chromatographed gave 120 mg. LIII and 5 mg. unidentified compound, m. 181-2° (MeOH). LXIII (248 mg.) in 15 ml. CC14 ozonized 8 min. at room temperature, CC14 removed in vacuo, the residue in 20 ml. Et2O and 10 ml. AcOH cooled in ice, treated with 2 ml. H2O and 100 mg. Zn dust, after 12 hrs. 20 ml. H2O added, the Et2O layer separated, combined with Et2O-washings of the aqueous layer, and worked up gave 5% LXIII, 40% XXVa, 30% Cl-free acidic fraction, and 4% 7-chloro-4,6-dimethoxy-2'-methylgris-2'-ene-3,4'-dione (LXIV), m. 167-8°, ν 1718 and 1683 cm.⁻¹, λ 323, 291, 235, 213 μ (log ϵ 3.70, 4.30, 4.19, 4.42). LXIII (137 mg.) in CC14 ozonized 0.5 hr. and worked up gave 10 mg. Cl-free solid, 30 mg. XXVa, 6 mg. LXIII, 1.2 mg. unidentified product, m. 143-9°, and 13 mg. Et 7-chloro-4,6-dimethoxy-3,4'-dioxogris-2'-ene-3'-carboxylate, m. 174° (EtOH), ν 1739, 1705, 1684, 1623, 1593 cm.⁻¹, not hydrolyzed by boiling 1-10N HCl, alkaline hydrolysis causing degradation. LXIV (10 mg.) in 1 ml. EtOAc added to a previously reduced catalyst prepared from 5 mg. PdCl2 and 20 mg. C in 1 ml. H2O, shaken 6 hrs. with H at room temperature and pressure (absorption complete in 3 hrs.), the recovered gum chromatographed in C6H6 on Al2O3, and the recovered gums from each eluate chromatographed on circular paper gave an alc. fraction (not investigated) and the racemate of 1,4-d-7-chloro-4,6-dimethoxy-2'-methylgrisan-3-one, d,d-7-chloro-4,6-dimethoxy-2'-methylgrisan-3,4'-dione, m. 170-5°.

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ACCESSION NUMBER: 1959:99795 CAPLUS

DOCUMENT NUMBER: 53:99795

ORIGINAL REFERENCE NO.: 53:18002e-i,18003a-c

TITLE: Transesterification of methyl methacrylate with amino alcohols. Preparation of a primary aminoalkyl methacrylate and 2-isopropenyl-4,4-dimethylloxazoline De Benneville, Peter L.; Luskin, Leo S.; Sims, Homer J.

CORPORATE SOURCE: Rohm & Haas Co., Philadelphia, PA

SOURCE: Journal of Organic Chemistry (1958), 23, 1355-7
CODEN: JOCEAH; ISSN: 0022-3263

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

OTHER SOURCE(S): CASREACT 53:99795

GI For diagram(s), see printed CA Issue.

AB (All nD and d. at 25°). Me2C(NH2)(CH2)3CHMe(CH2)2CH2OH (57.6 g.), 66.6 g. CH2:CMcCO2Me (I), 2 g. Al(OPr-iso)3, and 7.2 g. di- β -naphthol (II) heated 10 hrs. under a 20-cm. Vigreux column and total reflux distillation head (23 g. distillate collected, about half between 65-70°, and the remainder as the temperature slowly rose to 98°) and the product fractionated gave 74 g. (crude) CH2:CMcCO2(CH2)3CHMe(CH2)3C(NH2)Me2 (III), b0.8 115-20°, nD 1.4570, d. 0.9215, ν 1724, 1641, and 1565 cm.⁻¹, giving a tacky polymer when heated 16 hrs. at 70° with 1% di-Me azoisobutyrate (IV). (Procedure A) Me2C(NH2)CH2OH (V) (89 g.), 220 g. I, and 10 g. II heated 1 hr. under a short column (to remove a small amount H2O), 1 g. Ti(OPr-iso)4 (VI) added, distillation continued 15 hrs. (more VI added after 4 and 11 hrs.), the pot temperature adjusted to keep the overhead temperature below 70° as long as possible, and the residue fractionated, the crude product fractions combined, treated with 25 g. EtNCO in 200 cc. pentane, and the 2 layers formed distilled twice gave 52.3 g. O.C(CMe: CH2):N.CMe2.CH2 (VII), b30 62-3°, from the upper layer, and from the lower layer 11 g. VII, b36, 65-6°, nD 1.4536-1.4540, d. 0.9181, neutralization equivalent 142. In a run of similar size, the pot temperature held below 110° by operation at 400 mm., the crude product (89.5 g., b34 59-62°) dried over Na, and distilled gave 60% VII, b. 149°, ν 1658 and 1613 cm.⁻¹, λ (isooctane) 225 μ (log ϵ 3.79) (inflection).

A slight excess of saturated aqueous Na picrate solution added to 1.39 g. VII

ml. 0.5N NCl, the precipitate collected, and dried at 50° gave 2.6 g. VII picrate, m. 115-16°. (a) VII (28 g.) shaken 48 hrs. with PtO₂ under 50 lb. H pressure, fresh PtO₂ added, shaking continued 24 hrs., the mixture filtered, distilled in vacuo, the distillate (17.4 g., b₂₉ 50-2°) dried over Na, and redistd. gave O. C (Pr-iso)-N.CMe₂.CH₂ (VIII), b₃₅ 56°, n_D 1.4218, d. 0.8797, v 1665 cm.⁻¹ and 1612 cm.⁻¹ (b) V (89 g.) added to 89 g. Me₂CHCO₂H, the salt heated 5 hrs. under a distillation head (a 2-phase distillate formed), the wet product layer separated, dried over Na, and distilled gave 93 g. VIII, b₂₈ 50-3°, n_D 1.4210, d. 0.8770, v 1665 cm.⁻¹, ultraviolet end absorption below 245 mμ (Procedure B). V (1 mole) and 1 mole CH₂:CMeCO₂H treated as above gave 60 g. impure VII, b₃₅ 64-7°, n_D 1.4492, neutralization equivalent 143. CH₂:CMeCOCl (IX) (104.5 g.) added with stirring at 10-15° to 185 g. V in 100 ml. C₆H₆ during 2.5 hrs., the mixture stirred 3 hrs., refrigerated overnight, filtered, the solids washed with C₆H₆, the combined filtrates evaporated, and the residual oil (169 g.) distilled in the presence of p-HOC₆H₄NHPh gave 140 g. CH₂:CMeCONHCMe₂CH₂OH (X), b_{0.15} 97°, n_D 1.4785, v 1660 and 1621 cm.⁻¹ (c) X (32 g.) heated 4 hrs. at 175-80° and the crude product separated gave 24.3 g. VII, n_D 1.4545. In the presence of a few drops concentrated H₂SO₄, the distillation was completed in 1.5 hrs. at 180-90° and the product (24.5 g.) dried over Na, and redistd. gave 67% VII, identical with the material obtained by transesterification. Me₂C(OH)CH₂NH₂ and IX treated as above, the solids dissolved in 100 ml. H₂O, the solution extracted with two 50-ml. portions CHCl₃, and the extract evaporated finally in vacuo gave 34.5 g. CH₂:CMeCO₂NHCH₂C(OH)Me₂, m. 70-5°, nondistillable. VII (1 g.) containing 0.01 g. IV heated 24 hrs. at 75° gave a polymer, soluble in H₂O and C₆H₆.

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ACCESSION NUMBER: 1955:15976 CAPLUS

DOCUMENT NUMBER: 49:15976

ORIGINAL REFERENCE NO.: 49:3137a-i,3138a-i,3139a-i,3140a-i,3141a-i,3142a-i,3143a-i,3144a-i,3145a-i,3146a-i,3147a-i,3148a-i,3149a-i,3150a-i,3151a-b

TITLE: Oxazoles and oxazolones

AUTHOR(S): Cornforth, J. W.; Clarke, H. T.; et al.

CORPORATE SOURCE: Oxford Univ.; Princeton Univ. Press

SOURCE: Chemistry of Penicillin (1949) 688-848

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

GI For diagram(s), see printed CA Issue.

AB OXAZOLE SECTION: New methods for constructing the oxazole ring have been devised and the behavior of functional groups elucidated. The synthesis of oxazoles and imidazoles from K β-hydroxy-α-(α-alkoxyalkylideneamino)acrylates is given. A mixture of 51.1 g. AmCN and 24.5 g. EtOH was kept with 19.2 g. dry HCl below 0° for 2 wk, decomposed with 74 g. K₂CO₃ in Et₂O and distilled. The crude AmC(OEt):NH (62.4 g.), b₁₁ 52-65°, was shaken with cold aqueous H₂NCH₂CO₂Et.HCl for 1 h. The upper layer was fractionated to yield Et α-ethoxycaprylideneaminoacetate (I), b_{0.5} 91°, saponified on gentle warming to AmCO₂Et. The corresponding Me α-methoxycaprylideneaminoacetate (Ia), b_{0.1} 74°, was similarly prepared. A solution of 0.85 g. K in 2.5 g. EtOH and 14 g. Et₂O was diluted to 50 mL. with Et₂O, cooled to -15° and treated with a similarly cooled mixture of 4.85 g. I and 3.2 g. HCO₂Et, yielding after 3 h. at -10°, 2.6 g. of hygroscopic needles of C₅H₁₁C(OEt):NC(CO₂Et):CHOK (II). The corresponding K Me β-hydroxy-α-(α-methoxycaprylideneamino) acrylate (IIa) was obtained in 3.2 g.-yield from 3.75 g. Ia. Treatment of 2.6 g. II and 1.25

g. DL-penicillamine in 5 cc. EtOH with alc.-HCl gave crystalline DL-N-caproylpenicillamine, m. 137-8°. Treatment of II with ethereal HCl produced Et 2-amyl-oxazole-4-carboxylate, b₀.07 99° (dinitrophenyl-hydrazone, m. 165-6°; amide, m. 152°) saponified to 2-amyl-oxazole-4-carboxylic acid, m. 92-3° (PhNH₂ salt, m.98.5-9.5°) readily decarboxylated to 2-amyl-oxazole, b. 172-3°; picrate, m. 84.5-5.5°. This general synthesis of 2-substituted oxazoles and their 4-carboxylic acids has been extended to Et 2-phenyloxazole-4-carboxylate, m. 69-70°, the corresponding acid, m. 209°, and carried through to the known 2-phenyloxazole. The method can be also applied to the synthesis of imidazoles. Treatment of I with aqueous NH₄OH gave 2-amylimidazole-4-carboxylic acid, m. 230° (decomposition); with MeNH₂.HCl or alc. H₂NCH₂CO₂Et.HCl, I produced, resp., Et 2-amyl-1-methylimidazole-4-carboxylate (III), m. 42-3°, and Et 2-amylimidazole-4-carboxylate-1-acetate (IIIa), m. 61°. Similarly, Ia gave Me 2-amyl-1-methylimidazole, m. 66.7°, and Me 2-amylimidazole-4-carboxylate-1-acetate, m. 107°. Hydrolysis of III and IIIa yielded 1-methyl-2-amylimidazole-4-carboxylic acid, m. 121-3°, and 2-amyl-4-carboxylimidazole-1-acetic acid, m. 132-4°. Starting from PhCH₂CN, Et 2-benzylimidazole-4-carboxylate-1-acetate, m. 111-2°, was likewise prepared, converted by treating with MeOH into a Me Et ester. On heating with aqueous NH₄OH and with PhNH₂, 2-amyl-oxazole-4-carboxylic acid was converted into 2-amylimidazole, m. 33-4° and 1-phenyl-2-amylimidazole, m. 143-4°. Synthesis of oxazoles by rearrangement of oxazolones. The Na salt of 2-benzyl-4-hydroxymethylene-5-oxazolone (2.7 g.) in 50 mL. absolute MeOH was treated with 5 mL. absolute Et₂O containing 0.38 g. HCl. The gummy product (2.28 g.) was taken up in 10 mL. absolute MeOH and heated for 30 min. with 6.2 mL. H₂O containing 0.42 g. NaOH. The residue on evaporation was dissolved in 10 mL. of iced H₂O, acidified with dilute HCl to pH 6.5 and extracted with Et₂O, yielding 700 mg. 2-benzyl-oxazole-4-carboxylic acid, m. 158°. On heating at 220°, crude 2-phenyl-4-(α -hydroxyethylidene)-5-oxazolone rearranged to 2-phenyl-5-methyloxazole (IV), m. 184-5° (decomposition). Similarly, on heating to 230°, Na 4-hydroxymethylene- α -amyl-5-oxazolone rearranged to 2-amyl-oxazole-4-carboxylic acid. Evaporation of 2-(1-pentenyl)-4-(hydroxymethylene)-5-oxazolone in NaOH and fusion of the residue at 250° under reduced pressure yielded 2-pentenyl-oxazole-4-carboxylic acid, m. 145-7°. Incidental syntheses of oxazole derivs. The action of PhSO₃Ag on Me thiobenzylpenaldate di-Et acetal produced colorless prisms of 2-benzyl-oxazole-4-carboxylic acid, m. 156-7° and the dehydration of Et α -benzylamino-acetoacetate gave Et 2-phenyl-5-methyloxazole-4-carboxylate, m. 51-2°, hydrolyzed to the acid, m. 180-1°, decarboxylated at 220° in the presence of a trace of CuO to IV. Thus a reaction known to succeed with α -acylamino ketones and carboxylic esters is extended to β -keto esters. The 2-substituted oxazoles and their 4-carboxylic acids and esters are feebly basic, readily oxidized by cold aqueous KMnO₄ but stable to Br in CCl₄. The ring opens on warming with 2,4-(O₂N)₂-C₆H₃NNNH₂ in 2N HCl with a tendency to formation of glyoxal osazone derivs. Rosenmund reduction of 2-amyl-oxazole-4-carboxylic acid chloride produced 2-amyl-oxazole-4-carboxaldehyde, b₈ 108° (2,4-dinitrophenylhydrazones, m. 172-3°), converted by warming with D-penicillamine-HCl in AcOH to the thiazolidine, devoid of antibiotic properties. From the corresponding Et ester, 2-benzyl-4-carboxyoxazole hydrazide, m. 81-3° and benzylamide, m. 121-2° were prepared. In attempts to synthesize the thiazolidine-oxazolone structure for penicillin, attention was directed to the preparation of 5-alkoxyoxazoles and many variations of the general method of dehydrating α -acylamino

esters with P205 were introduced. By the use of PC15, P205, POC13, SOC12, and PhSO2Cl, the following new oxazoles were prepared (substituent given): 2-Ph, 5-MeO, b9 141°; 2-Ph, 5-PhCH2O, m. 56°; 2-PhCH2, 5-EtO, b15 152-4°; 2-PhCH2, 5-MeO, m.31-2°; 2-Am, 5-EtO, b0.8 82-5°; 2-Am, 5-MeO, b1.0 60-65°; 2-(1-C5H9), 5-EtO, b20 125-8° (C5H9 = pentenyl); 2-(1-C5H9), 5-MeO, b15 108-10°; 2-PhCH:CH, 5-EtO, m. 35°; 2-PhCH:CH, 5-Ph CH2O, picrate, m. 135° (decomposition); 2-Ph, 4-Me, 5-EtO, b10 151°; 2-Ph, 4-Me, 5-PhCH2O, picrate, m. 112-13°; 2-PhCH2, 4-Me, 5-EtO, b15 145-50°; 2-Am, 4-Me, 5-EtO, b3 92°; 2,4-Ph2, 5-EtO, m. 47-8°; 2-Ph, 4-PhCH2, 5-EtO, picrate, m. 105°; 2-Ph, 4-PhCH2, 5-PhCH2O, picrate, m. 117°; 2,4-(PhCH2)2, 5-EtO, b0.3 145-50°; 2-Am, 4-PhCH:CH, 5-EtO, m. 92°; 2-Ph, 4-CO2Et, 5-EtO, m. 75°; 2-Am, 4-CO2Et, 5-EtO, b0.1 122-5°; 2-(1-C5H9), 4-CO2Et, 5-EtO, b0.2 125°; 2-PhCH2, 4-CO2Et, 5-EtO, b0.1 165°. The possibility of converting an alkoxyoxazole to the corresponding oxazolone was realized by the catalytic hydrogenation of 2 g. of 2-phenyl-5-benzyloxyoxazole in 30 mL. dry dioxane in the presence of Pd-black to 2-phenyl-5-oxazolone, m. 91°. The converse reaction, transformation of an oxazolone to an alkoxyoxazole, has also been achieved. Methylation of 3 g. of 2-phenyl-4-carbethoxy-5-oxazolone with 500 mg. CH2N2 in 50 mL. Et2O yielded 2-phenyl-4-carbethoxy-5-methoxyoxazole, m. 72°. Similarly, methylation of 2-phenyl-4-carbomethoxy-2-oxazolin-5-one gave 2-phenyl-4-carbomethoxy-5-methoxyoxazole, m. 98°, identical with that prepared by the dehydration of BzNHCH(CO2Me)2 with PC15 in CC14. Attempts to obtain 5-alkoxyoxazole-4-carboxaldehydes covered a wide range. Formylation of BzNHCH2CO2Et and condensation with PhCH2NH2 in Et2O gave Et β-benzylamino-α-benzamidoacrylate, R'NHCH:C(CO2Et)NHCH2 (V; R = Ph, R' = PhCH2), m. 108°, cyclized by PBr3, POC13 or PC15 to 2-phenyl-4-benzylaminomethylene-5-oxazolone (VI), m. 134-7; Ac derivative, m. 140°. In the same way, Et β-benzylamino-α-phenylacetamido acrylate (VIa) with PBr3 gave 2-benzyl-4-benzylaminomethylene-5-oxazolone (VIb). Dehydration of Et α-benzamido-β,β-diethoxypropionate with PC15-POC13 yielded 2-phenyl-4-(ethoxymethylene)5-oxazolone (VII). Distillation of benzyl α-benzamido-β,β-diethoxypropionate gave a mixture of products including benzyl α-benzamido-β-ethoxyacrylate, m. 108-10°; benzyl 2-phenyloxazole-4-carboxylate, m. 106-7°; and VII. Attempts were made to cyclize α-benzyl-β-methyl-DL-phenylpenicilloate, HN.CH(CO2R').CMe2.S.CHCH(NHCOR)CO2CH2Ph (VIII, R = Ph, R' = Me) (VIIIa), m. 130°; dibenzyl-DL-phenylpenicilloate (VIII, R = Ph, R' = PhCH2) (VIIIb), m. 107-8°; and DL-2-(carboxy-1-hexenylaminomethyl)-5,5-dimethyl-4-carbomethoxythiazolidine benzyl ester (VIII, R = 1-pentenyl, R' = Me) (VIIIc). The action of PC15 on VIII and VIIIa gave definite evidence of formation of thiazolidinylalkoxyoxazoles and cyclization of VIIb and chromatog. purification of the product gave benzyl 2-(2-phenyl-5-benzyloxy-4-oxazolyl)-5,5-dimethylthiazolidine-4-carboxylate, m. 120-5°, absorption band at 2850 Å. This reduced in EtOAc using a Pd-BaSO4 catalyst with 2 mol H, corresponding to removal of 2 PhCH2 groups, yielded a product with no-antibiotic activity. The simpler thiazolidines were also investigated. The reaction of 3-methyl-2-(benzamidocarbethoxymethyl)-thiazolidine with PC15 gave a Cl-containing product, converted by NaHCO3 to a probable sulfoxide. With PC13, a product was obtained, which was converted by aqueous KOH to 2-phenyl-4-hydroxymethylene-5-oxazolone. β-Methylaminoethyl mercaptan-HI (from 15 g. of 2-methylthiazoline-MeI) in 20 mL. H2O was treated with 11 g. of crude Na salt of C,N-diformylglycine Et ester and neutralized with AcOH. After 15 h., NaHCO3 was added and the dried CHCl3 exts. (120 mL.) were concentrated to give 6.55 g. of crude product, converted

by

treatment with 65.5 mL. of 10% HCl in EtOH to 4.4 g. of

2-(aminocarbethoxymethyl)-3-methylthiazolidine-2HCl (IX), m. 169-70° (decomposition). IX (10.0 g.) in 36.1 mL. of 2N NaOH and 35 mL. EtOH was stirred with 6.6 g. PhCH₂CS₂Me for 45 h., yielding 6.2 g. of colorless prisms of 2-[(phenylthioacetamido)carbethoxymethyl]-3-methylthiazolidine (X), m. 100-100.5°. Addition of 5.0 g X in 20 mL. CHCl₃ to 8.6 g. PhSO₃Ag and 2.5 mL. pyridine in 70-mL. CHCl₃ gave no identifiable organic products. The action of PhSO₃Ag on Me α -phenylthioacetamido- β , β -diethoxypropionate yielded a product from which Me-benzylpenaldate and 2-benzylloxazole-4-carboxylic acid were isolated. By the PC15 method it has been possible to prepare 4-(2-thiazolyl)-2-benzyl-5-ethoxyoxazole and 2-(p-nitrophenyl)-4-(5,5-dimethyl-4-carbomethoxy-2-thiazolyl)-5-ethoxyoxazole. Attempts to introduce a CHO group into the 4-position of 2-phenyl-5-ethoxyoxazole (XI) using PhNMeCHO and POCl₃ gave 2-phenyl-4-anilinomethylene-5-oxazoline. With AcNHBr, XI gave 2-phenyl-4-bromo-5-ethoxyoxazole, b_{0.8} 128°. The oxidation of 2-phenyl-4-methyl-5-ethoxyoxazole with SeO₂, CrO₃ or CrO₂Cl₂ resulted only in far-reaching breakdown. Condensation of PhCH₂CH₂COCO₂H with AcNH₂ or AmCONH₂ gave α -acetamido- and α -caproyl-amino- γ -phenylisocrotonic acid (XII). Treatment of the Et ester of XII with PC15 afforded 2-amyl-4-styryl-5-ethoxyoxazole (XIII), disrupted by ozonization with production of BzOH and H₂NCOCO₂Et. XIII (5.7 g.) in 100 mL. glacial AcOH was stirred with 9.0 g. of Pb(OAc)₄ for 3 h., yielding 6.1 g. of 2-(1-acetoxyamyl)-4-styryl-5-ethoxyoxazole, m. 90-1°, degraded by distillation with loss of AcOH to 2-(1-pentenyl)-4-styryl-5-ethoxyoxazole (XIV), m. 100°, reduced catalytically to XIII. Oxidation of 2.83 g. XIV in 30 mL. tert-BuOH containing 0.75 g. H₂O₂ and 30 mg. OsO₄ at 40-50° for 2 h. produced PrCHO and 5-ethoxy-4-styryl-oxazole-2-carboxaldehyde, m. 130.5°, converted into the thiazolidine, m. 169°, using DL-penicillamine. Cyclization of AmCONHCH(CO₂Et)₂ in dry alc. free CHCl₃ with PC15, yielded 2-amyl-5-ethoxyoxazole-4-carboxylic acid (XIV), m. 63.4°, which on refluxing with PC15 in CHCl₃ gave Et 2-amyl-5-chlorooxazole-4-carboxylate (XV), b_{0.3} 106°, catalytically reduced over Pd-BaSO₄ in xylene to 2-amyl-oxazole-4-carboxylate, acidified to the free acid (XVa), m. 93-4°, converted by alc. EtONa to XIV. Treatment of 2 g. XVa with 1.09 g. PC15 in 10 mL. CHCl₃ and distillation produced the corresponding acid chloride, b_{0.3} 96°, converted by (NH₄)₂CO₃ in aqueous NH₄OH to the amide, m. 90°, which, distilled with P₂O₅, gave 2-amyl-5-chloro-4-cyanooxazole (XVb), b_{0.15} 72°. Reduction of 3.0 g. XVb in a suspension of 5.7 g. anhydrous SnCl₂ in 40 mL. dry ether yielded unstable 2-amyl-5-chloro-oxazole-4-carboxaldehyde (XVI) (dinitrophenylhydrazones, m. 109-10°), rearranging in 3 days at room temperature or on low pressure distillation to 2-amyl-oxazole-4-carboxylic acid chloride. Despite its instability, XVI readily combined with D-penicillamine-HCl to produce D-2-(2-amyl-5-chloro-4-oxazolyl)-5,5-dimethylthiazolidine-4-carboxylic acid-HCl, m. 150-2° (decomposition). A similar series of compds. starting with Et 2-phenyl-5-ethoxyoxazole-4-carboxylate (XVII) and proceeding to the thiazolidine was later prepared. XVII was saponified to the crystalline acid (XVIIa), m. 148°, converted to the acid chloride (XVIIb), m. 105-6°, and to Et 2-phenyl-5-chlorooxazole-4-carboxylate, m. 68°, by refluxing in xylene for 1 h. The corresponding acid (XVIIII), m. 178-4° (decomposition), was converted through the acid chloride, m. 118-20°, the amide, m. 183°, and the cyano compound, m. 112°, to 2-phenyl-5-chlorooxazole-4-carboxaldehyde (XIX), m. 91-3°. The addition of 1.14 g. aldehyde in 5 mL. EtOH and 10 mL. Et₂O to 0.93 g. D-penicillamine-HCl in 5 mL. H₂O and 0.65 g. AcONa, and passage of HCl through a filtered ethereal solution of the reaction product, yielded 1.5 g. of 2-(2-phenyl-5-chloro-4-oxazolyl)-5,5-dimethylthiazolidine-4-carboxylic acid-HCl, m. 178° (decomposition); Me ester-HCl, m. 120-2°; free acid, m. 166°; Me ester, m. 154°; PhCH₂ ester, m. 116-7°. The thiazolidine exhibited a

low order of antibiotic activity. A similar series of 2-benzoyloxazole derivs. have been prepared but the corresponding thiazolidine was inactive: 2-benzyl-5-ethoxy-oxazole-4-carboxylic acid, m. 118° (decomposition); Et ester, b0.1 165°; acid chloride, m. 81-2°; 2-benzyl-5-chlorooxazole-4-carboxylic acid, m. 183° (decomposition); Et ester, b0.02 170-5°; acid chloride, m. 156-7°; cyano compound, m. 49-50°; aldehyde [dinitrophenylhydrazone, m. 173°; semicarbazone, m. 185° (decomposition)]; 2-(2-benzyl-5-chloro-4-oxazoly1)-5,5-dimethylthiazolidine-4-carboxylic acid-HCl, m. 176-7° (decomposition). By refluxing 223 mg. XVIII in 3 mL. EtOH with 40 mg. Na, the Cl was replaced by the EtO group with formation of the corresponding acid, XVIIa. Distillation of the aldehyde XIX

at

0.1 mm. gave 2-phenyloxazole-4-carboxylic acid chloride, m. 107-8°, transformed by stirring with cold concentrated aqueous NH4OH to the amide. Similarly the acid chloride XVIIb was converted to the amide, m. 118-19°, rearranged by heating for a few min. at 140° to Et 2-phenyl-5-aminooxazole-4-carboxylate, m. 183deg;. All oxazoles found to undergo rearrangement may be formulated as 5-substituted oxazoles having a CO group in the 4-position, the general case being N:CR'.O.CR3:CCOR2 → N:CR'.O.CR2:CCOR3. Known examples of rearrangement are tabulated. Since the mol. is unstable when R3 and R2 are Et and Cl, resp., or when R3 and R2 are Cl and H, resp., it is deduced that the ethoxy aldehydes should show too great stability for successful synthesis. Cyclization of AmCONHCHCNCO2Et with P2O5 in CHCl3 gave 2-amyl-4-cyano-5-ethoxyoxazole, b0.03 98°, not reduced to the aldehyde by SnCl2 in Et2O. No 4-acetyloxazole was obtained from the MeMgI reaction product but the isolation of Et α-caproylaminoacetate (dinitrophenylhydrazone, m. 166-7°) indicated oxazole ring cleavage. The dehydration of 2-phenyl-5-ethoxyoxazole-4-carboxamide with POC13 or the ethylation with MeCHN2 of the crude oxazoles obtained by treating BzNHCHCNCO2H with Ac2O produced 2-phenyl-4-cyano-5-ethoxyoxazole, m. 77°. The previously unknown 5-aminooxazoles were prepared thus: treatment of 7 g. BzNHCH(CN)CO2Et, m. 138°, in 125 mL. CHCl3 with 6.2 g. PC15 gave 4.5 g. Et 2-phenyl-5-aminooxazole-4-carboxylate, m. 185°, also prepared by the action of POC13 on Bz-NHCH(CONH2)CO2Et. Condensation of 1.18 g. H2NCH-(CO2Et)2 with 1.13 g. PhNHOEt by heating for 30 min. at 110° gave the alternative compound, formulated as 2-phenyl-4-carbethoxy-5-imidazolone, m. 275°. Similarly were prepared Et 2-benzyl-5-aminooxazole-4-carboxylate (XX), m. 124° and the corresponding 2-benzyl-4-carbethoxy-5-imidazolone, m. 254° (decomposition); 2-(1-pentenyl)-4-carbethoxy-5-aminooxazole, m. 105°; 2-amyl-4-carbethoxy-5-aminooxazole (XXa), m. 104° and the corresponding 2-amyl-4-carbethoxy-5-imidazolone, m. 230° (decomposition). On heating at 170° for 5 min., XXa was entirely converted into AmCONHCH(CN)CO2Et, m. 83°. Heating either XX or PhCH2CONHCH(CN)CO2Et at 160-70° for 15 min. produced an equilibrium mixture with the open chain ester predominating. This same mixture was formed by heating 2-benzyl-5-ethoxyoxazole-5-carboxylic amide, probably through initial rearrangement to the aminooxazole. Stirring 35 g. NCCH2CO2CH2Ph in 40 mL. of chilled glacial AcOH with saturated aqueous NaNO2 (16.5 g.)

yielded

29 g. NCC(NOH)CO2CH2Ph, m. 119°, reduced with Al-Hg to NCC(NH2)CO2CH2Ph, m. 95°, and benzoylated to NCCH(NHBz)CO2CH2Ph, m. 130°, converted by heating at 160° for 5 min. to 2-phenyl-4-carbobenzyloxy-5-aminooxazole, m. 203°. The 4-carbethoxy-5-aminooxazoles are feebly basic substances whose HCl salts dissociate readily. XXa.HCl, on boiling with ethereal EtOH gave AmCONHCH(CONH2)CO2Et, m. 150-1°, along with NH4Cl. Treatment of 1 g. XXa in 10 mL. dry Et2O at -15° with NOCl gave a low yield of Et 2-amyl-oxazole-4-carboxylate, m. 92-3°. Formylation of 15 g. BzNHCH2CN in 200 mL. HCO2Et and 100 mL. benzene by addition of NaOEt (from 2.16 g. Na) in 100 mL. benzene produced, after treatment of the

intermediate $\text{BzNHC}(:\text{CHONa})\text{CO}_2\text{H}$ with dilute H_2SO_4 to pH 4, 2-phenyl-5-aminooxazole-4-carboxaldehyde (XXI), m. 172-3°, probably in the tautomeric form. Formylation of $\text{AmCONHCH}_2\text{CN}$ and distillation of the product yielded 2-amyl-oxazole-4-carboxylic acid amide, m. 154-5°, evidently by rearrangement of XXI. The action of POCl_3 on $\text{Bz-NHCH}(\text{CONH}_2)_2$ and $\text{AmCONHCH}(\text{CONH}_2)_2$, m. 231°, gave 2-phenyl-5-amino-4-cyanooxazole, m. 233° (Ac derivative, m. 202-3°), and 2-amyl-5-amino-4-cyanooxazole, m. 117°. These aminooxazoles could not be reduced to aldehydes.

Saturation of 0.52 g. $\text{PhCH}_2\text{CSNHCH}(\text{CN})\text{CO}_2\text{Et}$, m. 157°, treated in 5 mL. dry EtOH with dry HCl at -10° and the solution evaporated after 12 h. at 20° in vacuo yielded 0.5 g. 2-benzyl-4-carbethoxy-5-aminothiazole, m. 180°. OXAZOLONE SECTION. Part. I. General Chemical of Oxazolones. Preparation of 2-Oxazolin-5-ones. The reaction of Ac2O with α -acylamino acids is the most general procedure by which new oxazolones, $\text{O.CR:N.CR}_1\text{R}_2.\text{CO}$, have been prepared (substituents given): 2-Me, 4-iso-Pr, b1.0 60°; 2-PhCH₂, 4-Me, b0.5-1.0 122-3°; 2-PhCH₂, 4-iso-Pr, b0.5 115-17°; 2,4-(PhCH₂)₂, oil; 2-Am, 4-PhCH₂, b5 135-8°; 2-(2-pentenyl), 4-PhCH₂, b1.0 155-7°; 2-PhCH₂, 4,4-Me₂ (I), m. 59.5°; 2-Ph, 4-iso-Bu, m. 56-7°; 2-PhCH₂, 4-sec-Bu, b2.0 137-9°; 2-Ph, 4,4-C₅H₁₀, m. 71°; 2-PhCH₂, 4-Me, 4-PhCH₂:CH, m. 56-7°; 2-Ph, 4-CO₂Et, m. 147-8°; 2-Am, 4-CO₂Et, oil; 2-Ph, 4-(p-MeOC₆H₄CH₂); 2-PhCH₂, 4-(p-MeOC₆H₄CH₂); and 2-PhCH₂, 4-iso-Bu. Similarly, heating 100 g. $\text{BzNHCH}_2\text{CO}_2\text{H}$ (II) in 300 mL. Ac₂O at 100° yielded 49 g. 2-phenyl-2-oxazolin-5-one (III), m. 94-5°, the only monosubstituted oxazolone prepared by this method. By warming $\text{BzNHCHPhCH}_2\text{CO}_2\text{H}$ in CHCl_3 with 1 equivalent of 2-benzyl-4-methyl-5-oxazolone, a good yield of 2-phenyl-4-benzyl-5-oxazolone, m. 68-9°, was obtained. Addition of 1 g. NaNO_2 in 20 mL. H₂O to 3 g. of $\text{BzNHC}(\text{CONHNH}_2):-\text{CHPh}$ in 30 mL. N HCl gave α -benzamidoacinnamic azide, m. 113-4° (decomposition), converted on boiling with EtOH or treatment with pyridine at room temperature

to

2-phenyl-4-benzylidene-5-oxazolone (IV). Similarly, $\text{Me}_2\text{C:C}(\text{NHBz})-\text{CON}_3$ was converted to 2-phenyl-4-isopropylidene-5-oxazolone (IVa). These type II (unsatd. substituent at the 4-position) unsatd. oxazolones are formed more readily than the above-listed type I (saturated substituent at the 4-position) saturated oxazolones to which the azide conversion could not be extended. Reduction of IV over Pd-C gave 2-phenyl-4-benzyl-5-oxazolone (V), m. 67-8°. IVa was similarly reduced in dioxane to give an oil which, treated with PhNH_2 in benzene, produced $\text{Me}_2\text{CCHCH}(\text{NHBz})\text{CONHPh}$, m. 211-2°. The possibility arose that any reagent capable of transforming an acid to its chloride might be expected to convert an α -acylamino acid to the corresponding oxazolone. Thus treatment of II in 15 mL. dioxane with 2 mL. PBr_3 gave III. Similarly, 14.5 g. $\text{PhCH}_2\text{CONHCMe}_2\text{CO}_2\text{H}$ in 150 mL. dioxane was treated with 18 g. PBr_3 . The solid product suspended in dioxane and treated with slight excess of CH_2N_2 in ether yielded I, converted by PhCH_2NH_2 into $\text{PhCH}_2\text{CONHCMe}_2\text{CONH}_2$, m. 122-3°. Treatment of $\text{PhCH}_2\text{CHNHBzCO}_2\text{H}$ in pyridine with PBr_3 likewise gave the known V. Attempts to prepare 2-benzyl-5-oxazolone from $\text{PhCH}_2\text{CONHCH}_2\text{CO}_2\text{H}$ gave an unstable oil, converted by PhCH_2NH_2 into $\text{PhCH}_2\text{CONHCH}_2\text{CONHCH}_2\text{Ph}$. Conversion of $\text{PhCH:C}(\text{NHBz})\text{CO}_2\text{H}$ into IV was effected by POCl_3 , SOCl_2 , pyridine, by ClCH_2COCl and K_2CO_3 , and by AcCl in dioxane. Oxazolones have been produced by treating $\text{PhCH}_2\text{OCOC}_1$ with acylamino acids. Apart from direct dehydration, three methods are known for the preparation of type II oxazolones; the Erlenmeyer aldehydeacylglycine synthesis, the Bergmann-Stein reaction of N-(α -haloacyl)amino acids with Ac₂O, and the dehydration of β -hydroxy- α -acylamino acids. In that III reacts with Me_2CO in the presence of NaOAc to yield IVa in the absence of Ac₂O, it is suggested that III is an intermediate in the Erlenmeyer synthesis. In the presence of a little pyridine, BzH condenses with III to produce IV. Similarly, 2-phenyl-4-propylidene-5-oxazolone, m. 88-9°, was obtained in good yield from III and EtCHO. By adding

Ac2O dropwise with stirring to 17.9 g. II and 6.1 g. fused NaOAc in 580 mL. Me2CO, refluxing for 3-4 h. at 59-62°, pouring the reaction mixture over 200 g. ice and diluting to 1500 mL. produced high yields (73%) of relatively pure 2-phenyl-4-isopropylidene-5-oxazolone, m. 98°.

Condensation of II with (EtO)2CHCHO and Ac2O gave 4,4'-glyoxalidenebis(2-phenyl-5-oxazolone), m. 325° (decomposition). Though no acyl interchange in the Erlenmeyer synthesis occurs with II, the formation of 2-methyl-4-benzylidene-5-oxazolone occurs when either PhCH2CONHCH2CO2H or AmCONHCH2CO2H (VI) is refluxed with BzH in the presence of Ac2O and NaOAc. Refluxing VI (15.1 g.) with 13.1 g. AmCO2Na and 61 g. (AmCO)2O in 49 mL. Me2CO for 24 h. at 75° gave α -caproyl-amino- β , β -dimethylacrylic acid, m. 162-3°, converted by melting and heating in vacuo at 180-90° into 2-amyl-4-isopropylidene-5-oxazolone, b0.03 60-2°. By Bergmann's method, 2-methyl-4-isopropylidene-5-oxazolone (VII) and 2-methyl-4-sec-butylidene-5-oxazolone were prepared from Me2CHCH2CH(NHCOCH2Cl)CO2H and EtMeCHCH(NHCOCH2Cl)CO2H. Carter's method was used to prepare VII by the action of Ac2O on Me2C(OMe)CHNH2CO2H. Ring opening Reactions of Oxazolones. The general reaction of oxazolones with H2O, ROH, RSH, NH3, RNH2 and RR'NH represented by $O \cdot CR_1N \cdot CR_2R_2 \cdot CO + HX \rightarrow OCHNCR_1R_2COX$, suggested originally the thiazolidine-oxazolone formulation of penicillin. Comparison of the reactivity of V with that of IV showed the former to be rapidly hydrolyzed by 2N aqueous acid or alkali under conditions not affecting the latter. V reacts with ROH more rapidly than III. In the presence of NaOMe or PhCH2NMe3-OH, IVa was converted quant. to Me2C:C(BzNH)CO2Me, m. 130-1°. The methanolysis of 2-benzyl-4-p-methoxybenzyl-5-oxazolone in dry absolute MeOH yielded (N-phenylacetyl-p-methoxyphenylalaninyl)-p-methoxyphenylalanine, m. 199-200°. The formation of the dipeptide may be due to an "ortho-ester" reaction with the imino-ether form of the oxazolone. Reaction of PhCH2SH with III and I yielded benzyl hippurate, m. 101-2° and Me2CHCH(NHCOCH2Ph)COSCH2Ph, m. 138.5°. Almost all types of oxazolones react with PhCH2NH2 to form α -acylaminoacyl-benzylamides. The reaction of V with d-MePhCHNH2 in dry dioxane was followed polarimetrically and at constant rotation, produced N-benzoylphenylalanine-d-N- α -phenylethylamide, m. 178-80°, $[\alpha]_D^{23}$ 28.5° (c 1, dioxane). The strongly enolized 2-phenyl-4-carbethoxy-5-oxazolone formed a salt with PhCH2NH2, converted on heating in xylene to the benzylamide, m. 132°. The reaction of PhNH2.HCl with III and 2-benzyl-4-sec-butyl-5-oxazolone gave the normal anilide and the corresponding acid. Reaction of V and 2-phenyl-4-isobutyl-5-oxazolone with L-HSCH2CH(NH2)CO2Me produced the normal amides, m. 128-9°, and 131-5°, resp., the NH2 group taking precedence over the SH group in the condensation. The action of N2H4 on oxazolones has been clarified. The addition of 18 g. -phenyl-4-methyl-5-oxazolone to excess 60% N2H4.H2O in EtOH and heating to 50-60° for 30 min. gave 17.5 g. benzoylalanine hydrazide, m. 142-4°; benzylidene derivative, m. 193-4°. Treatment of IV with N2H4.H2O also gave the normal hydrazide, PhCH:C(NHBz)CONHNH2, m. 113-14°, converted by heating the corresponding azide in xylene to 2-oxo-4-benzylidene-6-phenyl-1,3,5-oxadiazine, m. 174° (decomposition). Conversion of Me2C:C(NHBz)CON3 similarly produced 2-oxo-4-isopropylidene-6-phenyl-1,3,5-oxadiazine, m. 166-8°. A mixture of 5 g. IV, 10 mL. N2H4.H2O and 3 mL. EtOH was refluxed for 30 min. yielding 4-benzamido-3-phenyl-5-pyrazolidone, m. 228-9°, identical with the product formed by refluxing PhCH:C(NHBz)CONHNH2 (VIII), m. 157-8°, which N2H4.H2O for 30 min. Similarly, the hydrazide Me2C:C(NHBz)CONHNH2, m. 192-4°, was converted into 3,3-dimethyl-4-benzoylamino-5-pyrazolidine, m. 106-8°. The hydrazide VIII was boiled in N NaOH and the sparingly soluble salt on acidification gave 6-hydroxy-5-benzyl-3-phenyl-1,2,4-triazine, m. 175-6°; Ac derivative, 187-8°. Oxidation of XIII with K3Fe(CN)6 produced N,N'-bis(α -benzoylaminocinnamoyl)hydrazine, m. 265°.

together with a substance, m. 186-7°, with the probable structure $\text{PhCH}:\text{C}(\text{OH})\cdot\text{NBz}:\text{C}(\text{:CHPh})\cdot\text{CH}(\text{OH})\cdot\text{NBz}$, forming $\text{PhCH}_2\text{CH}(\text{NHBz})-(\text{CO}_2\text{H})$ on alkaline hydrolysis. REACTIONS OF TYPE II OXAZOLONES: Some reactions involving the double bond in type II oxazolones have been discovered. Treatment of IV in dry dioxane with 2 mol CH_2N_2 in dry Et₂O at 0° and allowing the solution to stand overnight at room temperature gave product, C₁₇H₁₃O₂N, m. 142-3°. Addition of liquid NH₃ to IVa with shaking and cooling in solid CO₂ gave a small yield of basic product, C₁₂H₁₇O₂N₃, m. 162-6°, probably by addition of 2 mol NH₃. Addition of H₂S and RSH to the double bond has been studied in connection with various syntheses of penicillamine. The addition, of 136 g. IVa in 675 mL. dry benzene to 3.38 g. Na in 675 mL. of chilled dry MeOH and 76.5 mL. PhCH₂SH produced Me₂CC(NHBz)CO₂Me, m. 137-8°, and Me₂C(SCH₂Ph)CH(NHBz)CO₂Me, m. 66-7°. The addition probably takes place after ring opening, since the oxazolone can be replaced by an acrylic ester. Similarly, IV under like conditions, gave PhCH(SCH₂Ph)CH(NHBz)CO₂Me, m. 164°. There is no evidence of direct addition of PhCH₂SH to the double bond. Addition of H₂S to IVa and VII in the presence of Et₃N yielded Me₂C(SH)CH(NHBz)COSH and Me₂C(SH)CH(NHAc)COSH, resp. The initial step is probably the addition of H₂S to the double bond. Anhydrous MeOH saturated with H₂S at 0° treated with IVa gave 2,5,5-trimethyl-2-thiazoline-4-carboxylic acid, b₂₅ 120°; picrate, m. 159°, probably formed by addition, followed by displacement. IV similarly yielded 2-phenyl-5,5-dimethyl-2-thiazoline-4-carboxylic acid, m. 124-6°. IVa was apparently converted by treatment with alc. NaSH to 2-phenyl-4-isopropylidene-5-thiazolone, m. 100.5-101.5°. The reactivity of the Me groups in IVa is sufficient to permit condensation reactions with BzH to produce 2-phenyl-4-benzylidenisopropylidene-5-oxazolone, m. 135°. A mixture of stereoisomers, m. 134-6°, was produced by heating a mixture of 35.8 g. BzNHCH₂CO₂H, 32 g. PhCH:CHAC, 15 g. of fused NaOAc and 50 mL. Ac₂O for 3 h. at 100°. IVa is a pseudo-acid and exhibits weak violet fluorescence in Et₃N. On addition of NaOMe to IVa in MeOH, the initial intense blue-violet fluorescence in UV light due to the presence of the propenyl-oxazole soon disappears with the formation of Me₂C:C(NHBz)CO₂Me by ring opening. Miscellaneous REACTIONS OF OXAZOLONES. Excess PhMgBr was added to 6.0 g. 2-phenyl-4-methyl-5-oxazolone in Et₂O and after refluxing for 6 h. the reaction product was hydrolyzed and extracted with Et₂O, yielding 4.6 g. 1,1-diphenyl-2-benzoylamino-propanol, m. 192-3°. With AgClO₄ in benzene, III in EtOH gave a complex, m. 146° (decomposition). A similar crystalline compound, m. 172° (decomposition) was formed with 2-benzyl-4-methyl-5-oxazolone (IX). Formylation of 2,4-diphenyl-5-oxazolone apparently produced a stabilized enolic form, PhC:N.CPh:COH.O, m. 110°. Oxidation of 2-phenyl-4-isobutyl- and 2-phenyl-4-benzyl-5-oxazolones with Hg(OAc)₂ gave the corresponding 4,4'-bisoxazolones, m. 138-42°, and 201-202.5°, resp. PSEUDO-OXAZOLONES. According to the method of Bergmann, 12 g. PhCHBr:CONHCH₂CO₂H was added to 5 mL. dry pyridine and 100 mL. Ac₂O and after 2.5 h. at 0° was poured over ice. The solid product was dried over NaOH and crystallized from warm MeOH by cooling to -50°, yielding 64% of 2-benzylidenepseudooxazolone (2-benzylidene-3-oxazolin-5-one), m. 92-4°, hydrolyzed by 0.5N HCl in acetone to PhCH₂-CONH₂, m. 153-7°. An attempt to prepare 2-benzyl-4-methylene-5-oxazolone by Bergmann's method from Ph-CHClCONHCHMeCO₂H gave the potent skin irritant 2-benzylidene-4-methylpseudo-5-oxazolone (X), m. 105-115°, hydrolyzed by aqueous acetone to PhCH₂CONH₂ and AcCO₂H, suggesting that the pseudooxazolones are intermediates in the Bergmann synthesis of type II oxazolones and that, in general, the latter are in dynamic equilibrium with the pseudooxazolones. In an attempt to use pseudooxazolones for the thiazolidine-oxazolone structure suggested for penicillin, Br was added to V and the product condensed with penicillamine (XI) in the presence of AcOK and AcOH. The low order of activity noted was probably due to

BrCH₂COCO₂H which has an activity of 6 units per mg. against Gram-pos. organisms. X (1 g.) in 40 mL. pure AcOEt was hydrogenated at several atmospheric

pressure in the presence of 2 g. active Raney Ni to IX, suggesting that the thiazolidine-oxazolone structure might be accessible by reduction of the corresponding pseudooxazolone. Ice-cold pyridine (20 mL.) in 65 mL. Me₂CO was mixed with 1 g. (EtO)₂CHCH(NHCOCHBrPh)CO₂H and after 3 h., the mixture was poured over crushed ice, extracted with CHCl₃, washed with aqueous NaHCO₃, dried by passage through acid-washed Al₂O₃, and the filtrate was evaporated, yielding 4.8 g. oily 2-benzylidene-4-(diethoxymethyl)pseudo-5-oxazolone, which failed to condense with XI. In another attempt, (EtO)₂CHCH(NHCOCHCIPh)CO₂Me was condensed with XI to give α-Me α-chlorobenzylpenicillate (XII). On treatment of crude XII (5.2 g.) with a mixture of 10.8 g. pyridine and 35.2 mL. Ac₂O with shaking and cooling, a dark brown gum was formed, which, crystallized from Et₂O at -50°, gave a "dehydropenicillin" (XIII), C₁₆H₁₆O₄N₂S, m. 90-5° (decomposition). Addnl. information in printed abstract

=> s oxazoline and polymerization inhibit?

10405 OXAZOLINE
376602 POLYMERIZATION
2157778 INHIBIT?
6011 POLYMERIZATION INHIBIT?
(POLYMERIZATION(W)INHIBIT?)

L9 7 OXAZOLINE AND POLYMERIZATION INHIBIT?

=> d 19 ibib abs 1-

YOU HAVE REQUESTED DATA FROM 7 ANSWERS - CONTINUE? Y/(N):y

L9 ANSWER 1 OF 7 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2006:1033648 CAPLUS
DOCUMENT NUMBER: 145:400523
TITLE: Solid additive composition and method thereof
INVENTOR(S): Burrington, James D.; George, Herman F.; Byfleet, William D.; Baker, Mark R.; Hurley, Susie; Sumiejski, James L.; Ineman, Jennifer M.
PATENT ASSIGNEE(S): The Lubrizol Corporation, USA
SOURCE: PCT Int. Appl., 22pp.
CODEN: PIXXD2
DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2006105025	A1	20061005	WO 2006-US11121	20060327
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, GR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KN, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
US 20060229215	A1	20061012	US 2005-92403	20050329
AU 2006230103	A1	20061005	AU 2006-230103	20060327
CA 2603007	A1	20061005	CA 2006-2603007	20060327

EP 1877527 A1 20080116 EP 2006-739737 20060327
 R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE,
 IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR
 JP 2008534746 T 20080828 JP 2008-504228 20060327
 IN 2007DN07685 A 20071109 IN 2007-DN7685 20071008
 CN 101184831 A 20080521 CN 2006-80018527 20071127
 PRIORITY APPLN. INFO.: US 2005-92403 A 20050329
 WO 2006-US11121 W 20060327

AB An additive composition comprises (A) an additive comprising a
 nitrogen-containing
 detergent, a fatty acid based friction modifier, or a mixture thereof and
 (B) a solid organic matrix material where the additive composition is a solid
 that
 melts in the range of .apprx.25 to .apprx.200°. A method to
 enhance performance of a functional fluid, especially a fuel composition or
 power
 transmission fluid of a vehicle powered by an internal combustion engine,
 comprises adding a performance-enhancing amount of the additive composition to
 the functional fluid.

REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS
 RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L9 ANSWER 2 OF 7 CAPLUS COPYRIGHT 2009 ACS on SIN

ACCESSION NUMBER: 2006:322508 CAPLUS
 DOCUMENT NUMBER: 144:351671
 TITLE: Optically clear stabilized polycarbonate-polyester
 compositions
 INVENTOR(S): Juikar, Vishvajit Chandrakant; Kannan, Ganesh; Wit,
 Gerrit De
 PATENT ASSIGNEE(S): General Electric Company, USA
 SOURCE: U.S. Pat. Appl. Publ., 16 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 20060074202	A1	20060406	US 2004-952613	20040929
WO 2006044087	A1	20060427	WO 2005-US33320	20050915
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KM, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
EP 1797136	A1	20070620	EP 2005-846423	20050915
R: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LI, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR				
CN 101065433	A	20071031	CN 2005-80040770	20050915
JP 2008514756	T	20080508	JP 2007-533564	20050915
PRIORITY APPLN. INFO.:			US 2004-952613 A 20040929	
			WO 2005-US33320 W 20050915	

AB A stabilized thermoplastic resin composition comprises structural units derived
 from at least one substituted or unsubstituted polycarbonate, at least one
 substituted or unsubstituted polyester, a cyclic iminoether-containing
 compound,

and an additive. In another embodiment, a stabilized thermoplastic resin composition comprises structural units derived from at least one substituted or unsubstituted polycarbonate, at least one substituted or unsubstituted polyester, a cyclic iminoether-containing compound, a quencher, and an additive.

The composition has improved optical properties, thermal properties and stability. Thus, a composition comprising Lexan polycarbonate (75%), cyclohexanedimethanol-ethylene glycol-terephthalic acid copolymer (25%), an oxazoline derivative (0.1% based on the polymer blend), and phosphoric acid (50 ppm) was extruded at 270°, a feed rate of 15 kg/h, and a screw speed of 300 rpm. The composition was pelletized, dried for at least 4 h at 100°, and injection molded at 280° to obtain a test sample showing a yellowness index of 1.56, a haze of 1.22%, and a transmission of 89.9% (ASTM D-1003).

L9 ANSWER 3 OF 7 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2005:472105 CAPLUS

DOCUMENT NUMBER: 143:8161

TITLE: Method for the esterification of alcohols with olefinically unsaturated carboxylic acids

INVENTOR(S): Glos, Martin

PATENT ASSIGNEE(S): Clariant G.m.b.H., Germany

SOURCE: PCT Int. Appl., 20 pp.

CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: German

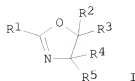
FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2005049544	A1	20050602	WO 2004-EP12790	20041111
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
DE 10354652	A1	20050707	DE 2003-10354652	20031122
AU 2004291298	A1	20050602	AU 2004-291298	20041111
CA 2546819	A1	20050602	CA 2004-2546819	20041111
EP 1687251	A1	20060809	EP 2004-797820	20041111
EP 1687251	B1	20090318		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK, IS				
BR 2004016734	A	20070116	BR 2004-16734	20041111
CN 1902155	A	20070124	CN 2004-80030595	20041111
JP 2007511561	T	20070510	JP 2006-540260	20041111
ES 2320227	T3	20090520	ES 2004-797820	20041111
MX 2006005674	A	20060817	MX 2006-5674	20060519
US 20070149803	A1	20070628	US 2006-580384	20060522
PRIORITY APPLN. INFO.:			DE 2003-10354652	A 20031122
			WO 2004-EP12790	W 20041111

OTHER SOURCE(S): MARPAT 143:8161

GI



AB The invention relates to a method for producing esters from alcs. and olefinically unsatd. carboxylic acids by reacting an alc. with an olefinically unsatd. carboxylic acid or a reactive derivative thereof, in the presence of between 1 ppm and 1 weight % oxazoline I in relation to the weight of the reaction mixture of alc. and olefinically unsatd. carboxylic acid/carboxylic acid derivative, R1, R2, R3, R4 and R5 representing hydrogen or branched, linear, cyclical, saturated or unsatd. hydrocarbon radicals containing up to 25 C atoms that can be substituted by heteroatoms, and R1, R2, R3, R4 and R5 being the same or different to prevent formation of polymer on surfaces in reactor.

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L9 ANSWER 4 OF 7 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:553220 CAPLUS

DOCUMENT NUMBER: 111:153220

ORIGINAL REFERENCE NO.: 111:25533a,25536a

TITLE: Purification of unsaturated carboxylic acid

isocyanatoalkyl esters by distillation

INVENTOR(S): Abe, Tetsuo; Yokoo, Hidejiro; Wakasa, Masami

PATENT ASSIGNEE(S): Showa Rodia Kagaku Co., Ltd., Japan

SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01042463	A	19890214	JP 1987-198157	19870810
JP 07103085	B	19951108		

PRIORITY APPLN. INFO.: JP 1987-198157 19870810

OTHER SOURCE(S): MARPAT 111:153220

AB The title esters, useful as monomers, are purified by distillation in the presence of ≥ 1 compound selected from phenothiazine (I), alkylphenols, hydroquinone, alkylhydroquinones, p-MeOC₆H₄OH, tannic acid, and anthraquinone and ≥ 1 compound selected from Et₂NCH₂CH₂OH (II), N-nitroso-N-arylhydroxylamine NH₄ salts, N-nitroso-N-propylurethane, H₂NNHCH₂CH₂OH, and C₆H₄(NO₂)₂ to prevent popcorn polymerization CH₂:CMeCO₂H

(320

g) was gradually added to mixture of 300 g 2-oxazolidinone, I, and toluene while bubbling with HCl over 60 min, and the reaction mixture was further stirred at 60° for 30 min, and then heated at 80° while bubbling with COCl₂. After distilling off toluene, 230 g reaction mixture containing CH₂:CMeCO₂CH₂CH₂NCO (III) was distilled with II under 10-12 mmHg

while

adding 50 g reaction mixture containing II dropwise to give 108 g III, vs. formation of polymers preventing distillation for a control without addition of II.

L9 ANSWER 5 OF 7 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:477514 CAPLUS

DOCUMENT NUMBER: 111:77514

ORIGINAL REFERENCE NO.: 111:13055a,13058a
 TITLE: Purification of unsaturated carboxylic acid isocyanatoalkyl esters by distillation
 INVENTOR(S): Abe, Tetsuo; Yokoo, Hidejiro; Nozawa, Kaneo
 PATENT ASSIGNEE(S): Showa Rodia Kagaku Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 6 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01042461	A	19890214	JP 1987-198155	19870810
JP 07049413	B	19950531		

PRIORITY APPLN. INFO.: JP 1987-198155 19870810
 OTHER SOURCE(S): MARPAT 111:77514

AB The title esters, useful as monomers, are purified by distillation under continuous or intermittent addition of nitrite esters in the presence of Sn(2+) and/or Fe(2+) compds. to prevent popcorn polymerization CH2:CMcCO2H
 (320

g) was gradually added to a solution of 300 g 2-oxazolidinone in toluene containing phenothiazine while bubbling with HCl at 60° over 60 min, and the reaction mixture was further bubbled with HCl at 60° for 30 min, and then heated at 80° while bubbling with COCl2. After distilling off toluene, 230 g product containing CH2:CMcCO2CH2CH2NCO (I) was distilled with SnCl2 and the HNO2 ester (II) of HOCH2CH2OCH2CH2O2Bu under dropwise addition of 50 g product containing II to give 115 g I. When the reaction product was distilled without addition of SnCl2 and II, granules of polymerized matter were formed at the upper part of the distillation tower and polymer beads grew in the reaction mixture

L9 ANSWER 6 OF 7 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1989:457070 CAPLUS
 DOCUMENT NUMBER: 111:57070
 ORIGINAL REFERENCE NO.: 111:9675a,9678a
 TITLE: Purification of unsaturated carboxylic acid isocyanatoalkyl esters by distillation
 INVENTOR(S): Abe, Tetsuo; Yokoo, Hidejiro
 PATENT ASSIGNEE(S): Showa Rodia Kagaku Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 6 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01042462	A	19890214	JP 1987-198156	19870810
JP 07049414	B	19950531		

PRIORITY APPLN. INFO.: JP 1987-198156 19870810
 OTHER SOURCE(S): MARPAT 111:57070

AB Unsaturated carboxylic acid isocyanatoalkyl esters, useful as monomers, are purified by distillation under continuous or intermittent feeding of cupferrons dissolved in glycols to prevent popcorn polymerization An aqueous solution of

150 g 2-isopropenyl-2-oxazoline, a solution of 200 g COCl2 in CH2Cl2, and an aqueous NaOH solution were simultaneously added to CH2Cl2 at ≤15°, and the reaction mixture was further stirred for several mins and separated After distilling off CH2Cl2 from the organic layer, the product

was distilled with cupferron (I) dissolved in ethylene glycol (II) under decreased pressure while continuously adding a solution of I in II to give 181 g CH₂:CMeCO₂CH₂CH₂NCO, vs. formation of polymerized matter preventing distillation for a control without addition of I.

L9 ANSWER 7 OF 7 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1961:105947 CAPLUS

DOCUMENT NUMBER: 55:105947

ORIGINAL REFERENCE NO.: 55:19960a-e

TITLE: 4,4-Dialkyl-2-vinyl- and
4,4-dialkyl-2-isopropenyloxazolines and
5,6-dihydro-4H-1,3-oxazines and their spirans
Luskin, Leo Samuel; De Benneville, Peter L.

INVENTOR(S):
PATENT ASSIGNEE(S): Rohm & Haas Co.

DOCUMENT TYPE: Patent

LANGUAGE: Unavailable

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
DE 1067437		19591022	DE	

AB Acryloyl halides or their α -alkyl derivs. were treated with excess alkanolamines, such as 2-amino-2-methylpropanol (I), 3-amino-3-methylbutanol (II), 2-amino-2-methyl-1-decanol (III), or 2-amino-2-methyl-1-octanol (IV) and the amides formed heated at 170-230° in the presence of a strong mineral acid and a polymerization inhibitor to give the title oxazolines or oxazines. The products were fungicides and could be polymerized or copolymerized. The fungicidal action was tested. The infrared spectra of the compds. prepared were discussed. E.g., I 105 and C6H6 50 was cooled at 10-15°, treated with CH₂:CMeCOCl (V) 104.5 in C6H6 100 parts over 2 hrs. at 10-40°, the mixture stirred 4 hrs. without cooling, the precipitate filtered off, washed with C6H6, the filtered solution and the washings combined, evaporated, and distilled in the presence of p-hydroxydiphenylamine

to give 90% N-(1,1-dimethyl-2-hydroxyethyl)methacrylamide, b0.15 92-105°, n25D 1.4785). The amide, p-hydroxydiphenylamine 0.5, and H₂SO₄ 0.5 was heated at 175-230° 1.5 hrs. and the mixture 24.5 parts of H₂O and oxazoline distilled simultaneously through a short column and redistd. to give 60% 2-isopropenyl-4,4-dimethyloxazoline, b24 58-9°, n25D 1.453, an effective fungicide and a monomer for polymerization and copolymerization. Similarly, II and V gave N-(1,1-dimethyl-3-hydroxypropyl)methacrylamide, b0.13 95-110°, and 2-isopropenyl-4,4-dimethyl-5,6-dihydro-4H-1,3-oxazine, b13 58-65°, n25D 1.4558. I and CH₂:CHCOCl gave oily N-(1,1-dimethyl-2-hydroxyethyl)acrylamide and 2-vinyl-4,4-dimethyloxazoline, b32 54°, n25D 1.4356). CH₂:CHCOCl and III gave oily N-(1-methyl-1-hexyl-2-hydroxyethyl)acrylamide and 2-vinyl-4-methyl-4-hexyloxazoline, b13 78-90°, n25D 1.4533). IV and V gave N-(1-methyl-1-octyl-2-hydroxyethyl)methacrylamide and 2-isopropenyl-4-methyl-4-octyloxazoline, b25 72-5°, n25D 1.4482. V and 1-amino-1-(hydroxymethyl)cyclohexane gave N-[1-(hydroxymethyl)cyclohexyl]methacrylamide and 2-isopropenyl-4-spiro(cyclohexanooxazolidine), b12 90-110°, n25D 1.4853, which could be used for treatment of leather and for polymerization.

=> E GLOS MARTIN/AU 25

E1 1 GLOS L/AU

E2 1 GLOS M/AU

E3 39 --> GLOS MARTIN/AU

E4 1 GLOS MERRILL/AU
 E5 2 GLOS MICHAEL/AU
 E6 3 GLOS MIROSLAV/AU
 E7 4 GLOS P/AU
 E8 1 GLOS RAINER/AU
 E9 1 GLOS S/AU
 E10 1 GLOS SABRINA/AU
 E11 5 GLOS STEFAN/AU
 E12 2 GLOS T/AU
 E13 2 GLOS U/AU
 E14 1 GLOSATTO C V L/AU
 E15 1 GLOSAUER EVA M/AU
 E16 11 GLOSAUER OTTO/AU
 E17 9 GLOSBY A A/AU
 E18 1 GLOSCH BERNADETT/AU
 E19 4 GLOSCH H/AU
 E20 1 GLOSCH HARTMUT/AU
 E21 3 GLOSCOW ELEANOR J/AU
 E22 6 GLOSE CHRISTOPHER R/AU
 E23 4 GLOSE CHRISTOPHER RAYMOND/AU
 E24 4 GLOSE KEN/AU
 E25 3 GLOSE KENNETH H/AU

=> S (E2 OR E3)
 1 "GLOS M"/AU
 39 "GLOS MARTIN"/AU
 L10 40 ("GLOS M"/AU OR "GLOS MARTIN"/AU)

=> s l4 and l10
 L11 2 L4 AND L10

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 YOU HAVE REQUESTED DATA FROM 2 ANSWERS - CONTINUE? Y/(N):y

L11 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2009 ACS on STN
 ACCESSION NUMBER: 2005:472105 CAPLUS
 DOCUMENT NUMBER: 143:8161
 TITLE: Method for the esterification of alcohols with
 olefinically unsaturated carboxylic acids
 Inventor(S): Glos, Martin
 PATENT ASSIGNEE(S): Clariant G.m.b.H., Germany
 SOURCE: PCT Int. Appl., 20 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: German
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

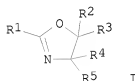
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2005049544	A1	20050602	WO 2004-EP12790	20041111
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
DE 10354652	A1	20050707	DE 2003-10354652	20031122

AU 2004291298	A1	20050602	AU 2004-291298	20041111
CA 2546819	A1	20050602	CA 2004-2546819	20041111
EP 1687251	A1	20060809	EP 2004-797820	20041111
EP 1687251	B1	20090318		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,				
IE, SI, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK, IS				
BR 2004016734	A	20070116	BR 2004-16734	20041111
CN 1902155	A	20070124	CN 2004-80030595	20041111
JF 2007511561	T	20070510	JP 2006-540260	20041111
ES 2320227	T3	20090520	ES 2004-797820	20041111
MX 2006005674	A	20060817	MX 2006-5674	20060519
US 20070149803	A1	20070628	US 2006-580384	20060522

PRIORITY APPLN. INFO.:

DE 2003-10354652	A	20031122
WO 2004-EP12790	W	20041111

OTHER SOURCE(S): MARPAT 143:8161
GI



AB The invention relates to a method for producing esters from alcs. and olefinically unsatd. carboxylic acids by reacting an alc. with an olefinically unsatd. carboxylic acid or a reactive derivative thereof, in the presence of between 1 ppm and 1 weight % oxazoline I in relation to the weight of the reaction mixture of alc. and olefinically unsatd. carboxylic acid/carboxylic acid derivative, R1, R2, R3, R4 and R5 representing hydrogen or branched, linear, cyclical, saturated or unsatd. hydrocarbon radicals containing up to 25 C atoms that can be substituted by heteroatoms, and R1, R2, R3, R4 and R5 being the same or different to prevent formation of polymer on surfaces in reactor.

REFERENCE COUNT: 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L11 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2009 ACS ON SIN

ACCESSION NUMBER: 2000:402279 CAPLUS

DOCUMENT NUMBER: 133:207839

TITLE: Aza-bis(oxazolines): New Chiral Ligands for Asymmetric Catalysis

AUTHOR(S): Glos, Martin; Reiser, Oliver

CORPORATE SOURCE: Institut fuer Organische Chemie, Universitaet

Regensburg, Regensburg, D-93040, Germany

SOURCE: Organic Letters (2000), 2(14), 2045-2048

CODEN: ORLEF7; ISSN: 1523-7060

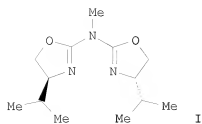
PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 133:207839

GI



AB Aza-bis(oxazolines), e.g. I, are introduced as chiral ligands for asym. catalysis combining the advantages of easy availability of bis(oxazolines) and backbone variability of aza-semicorrins. The title ligands could be attached to a polymeric support, which allowed for the development of easily recoverable copper(I)-catalysts for asym. cyclopropanation reactions.

REFERENCE COUNT: 71 THERE ARE 71 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT